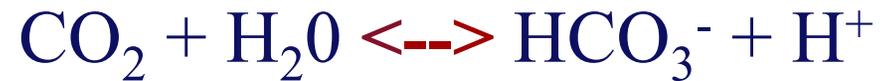


# **Ocean Acidification**

## **And Animal Physiology**



$$\text{pH} = \text{pK} + \log \frac{[\text{HCO}_3^-]}{[\text{CO}_2]}$$

# Outline for Today's Lecture

## 1. Introduction to acid-base physiology

- Buffering capacity and Ion transport
- Correlation with metabolic rate

## 2. What processes (and organisms) are sensitive?

- enzyme-mediated processes
- blood-oxygen binding
- Metabolic rate as CO<sub>2</sub>-sensitive indicator?

## 3. *Dosidicus gigas* as a canary in the coalmine.

## 4. Does a broadly-applicable critical CO<sub>2</sub> level exist?

# Organisms can control intracellular $\text{PCO}_2$ and pH

QuickTime™ and a  
decompressor  
are needed to see this picture.

QuickTime™ and a  
decompressor  
are needed to see this picture.

QuickTime™ and a  
decompressor  
are needed to see this picture.

QuickTime™ and a  
decompressor  
are needed to see this picture.

QuickTime™ and a  
decompressor  
are needed to see this picture.

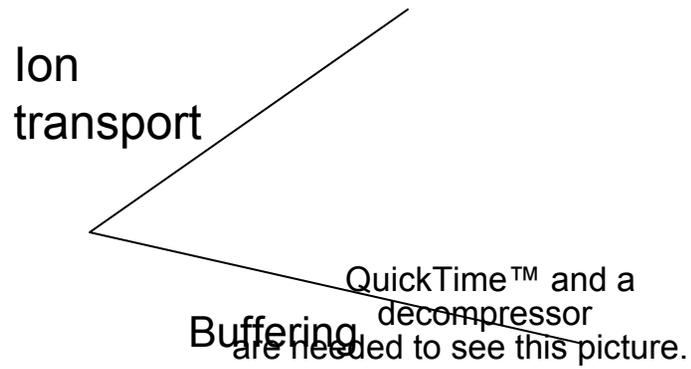
QuickTime™ and a  
decompressor  
are needed to see this picture.

QuickTime™ and a  
decompressor  
are needed to see this picture.

QuickTime™ and a  
decompressor  
are needed to see this picture.

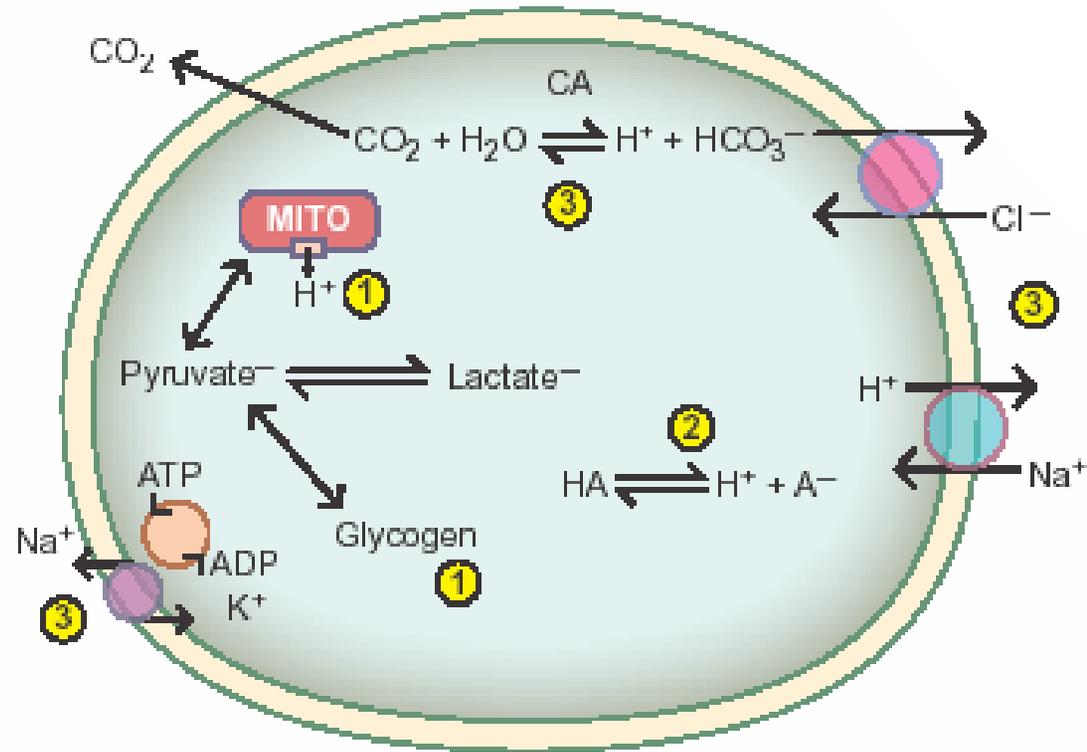
QuickTime™ and a  
decompressor  
are needed to see this picture.

Buffering moderates pH imbalance on short term (minutes), ion transport compensates on longer time scales (hours-days).



# Potential Impacts of CO<sub>2</sub>

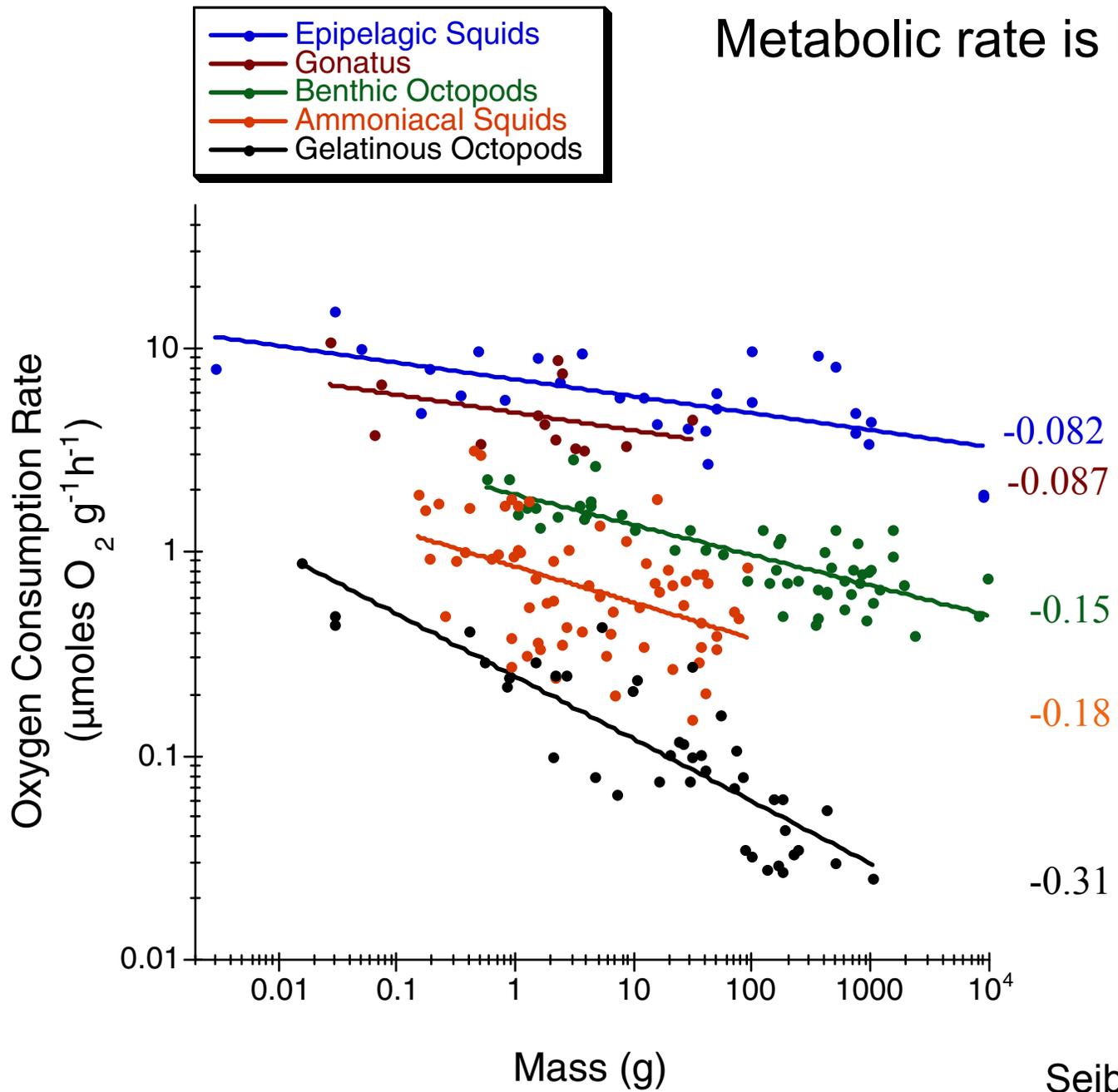
Brad A. Seibel and Patrick J. Walsh



Organisms capacity to control intracellular pH is an evolved function of the rate of production of acid-base equivalents (i.e. Dependent on metabolic rate).

Melzner et al., 2009. Biogeosciences

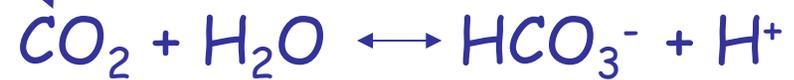
Metabolic rate is highly variable



# Combating Acidosis

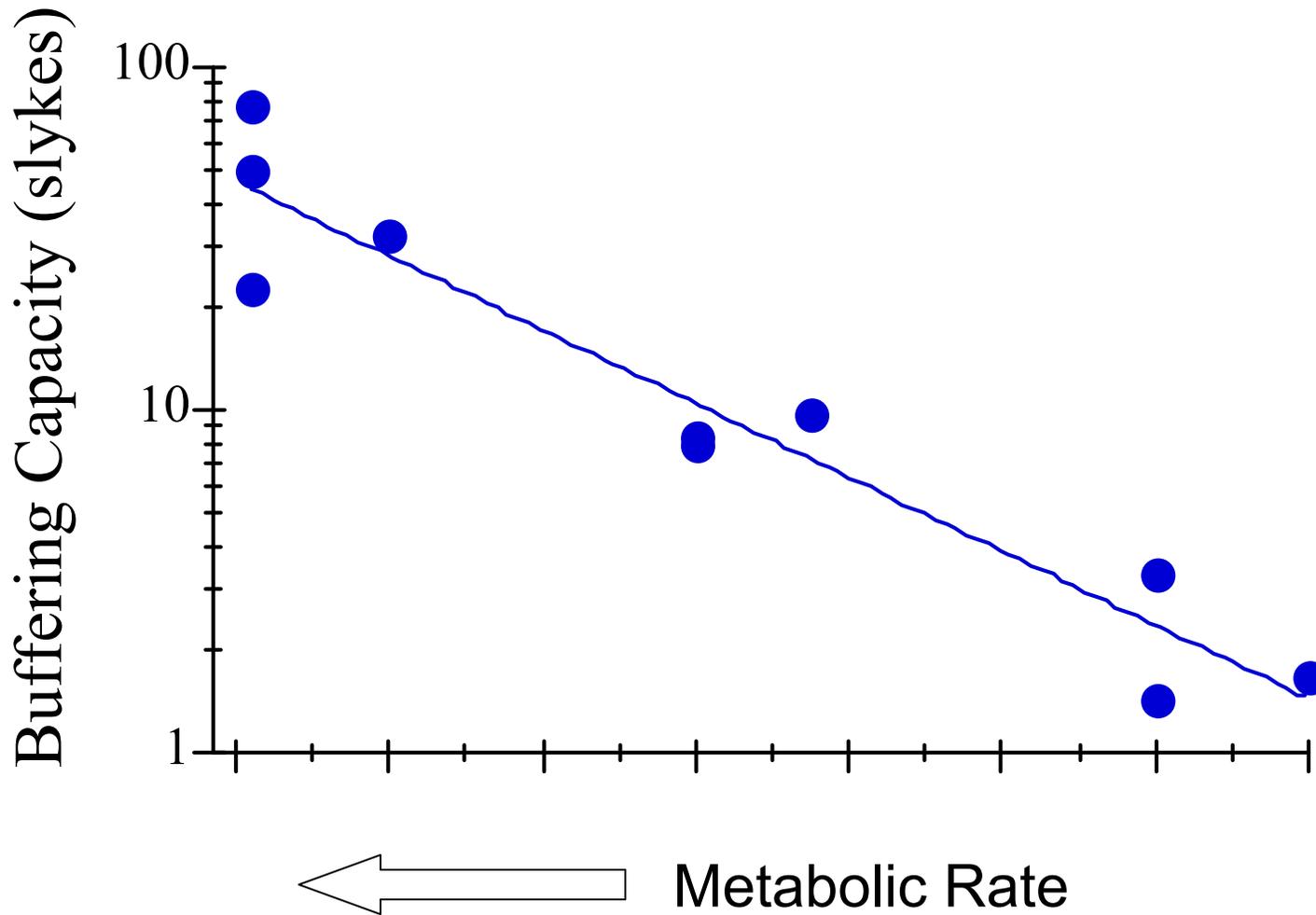
## 1. Buffering

$\text{CO}_2$

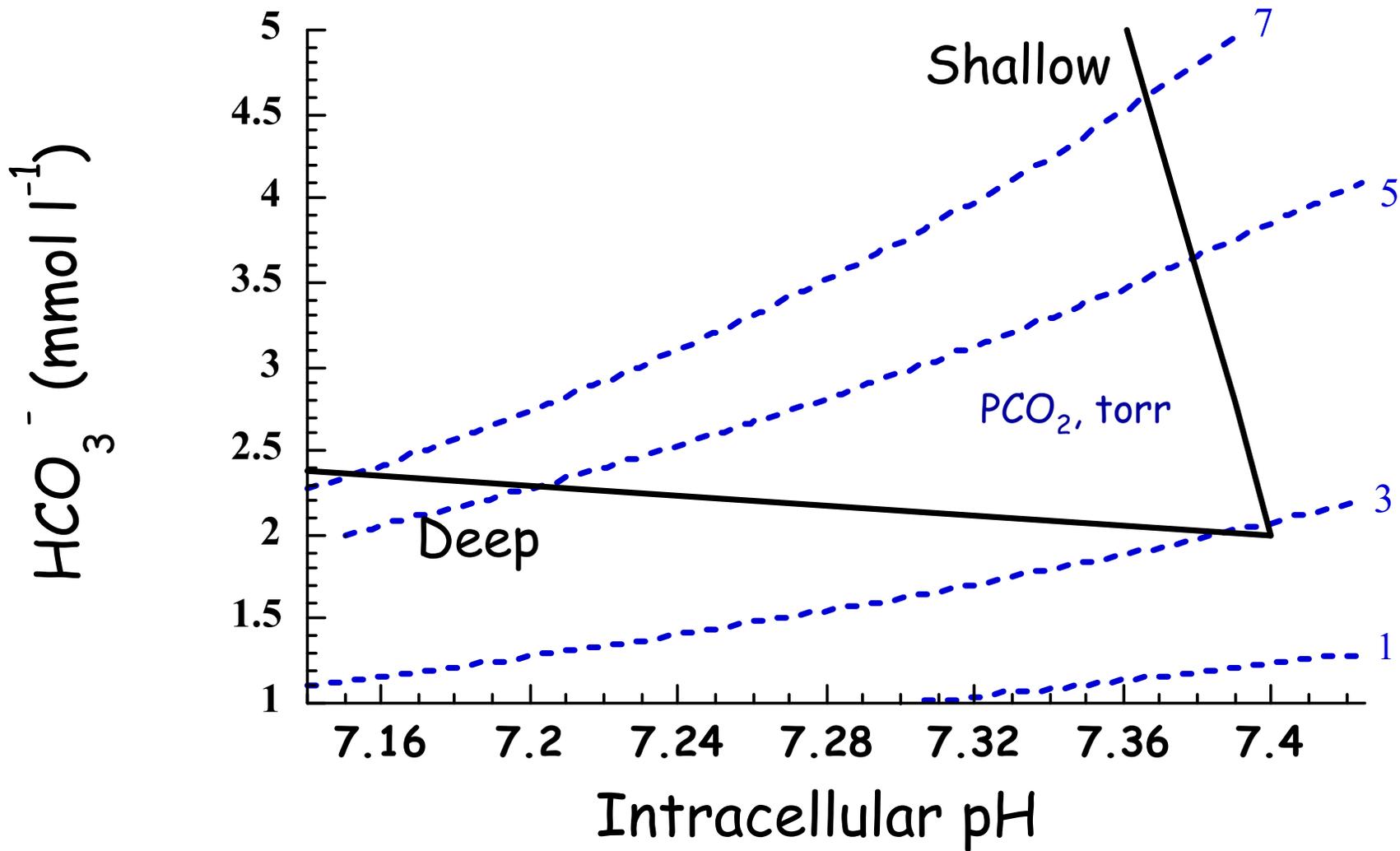


1

# Buffering Capacity is correlated with metabolic rate



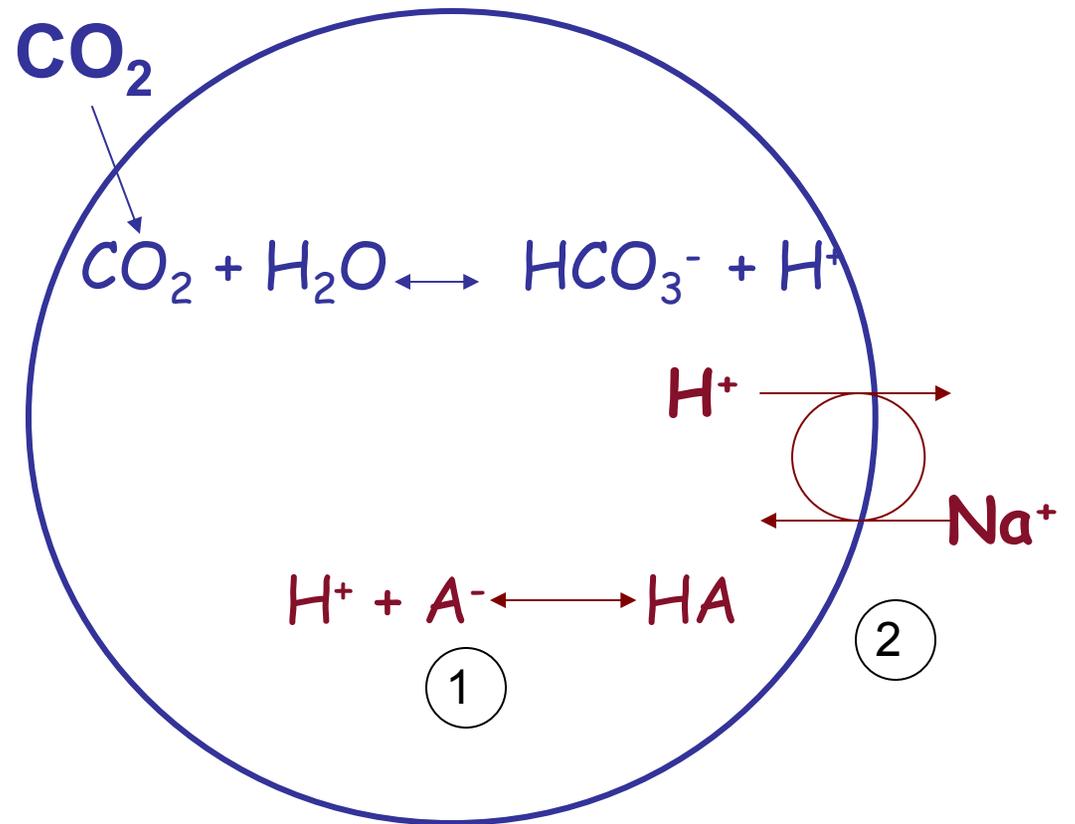
$$\text{pH} = \text{pK} + \log \frac{[\text{HCO}_3^-]}{\text{CO}_2}$$



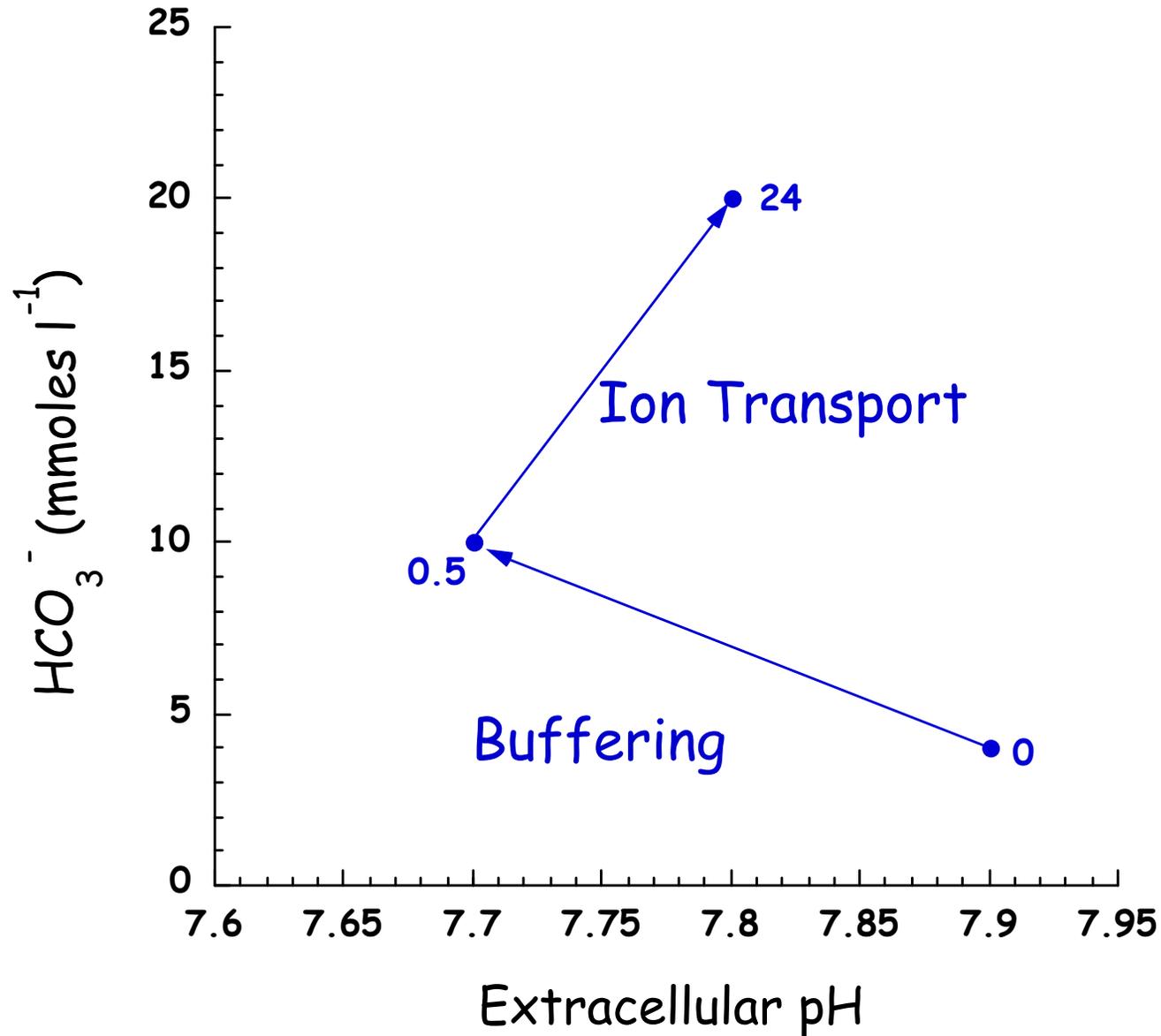
# Combating Acidosis

1. Buffering

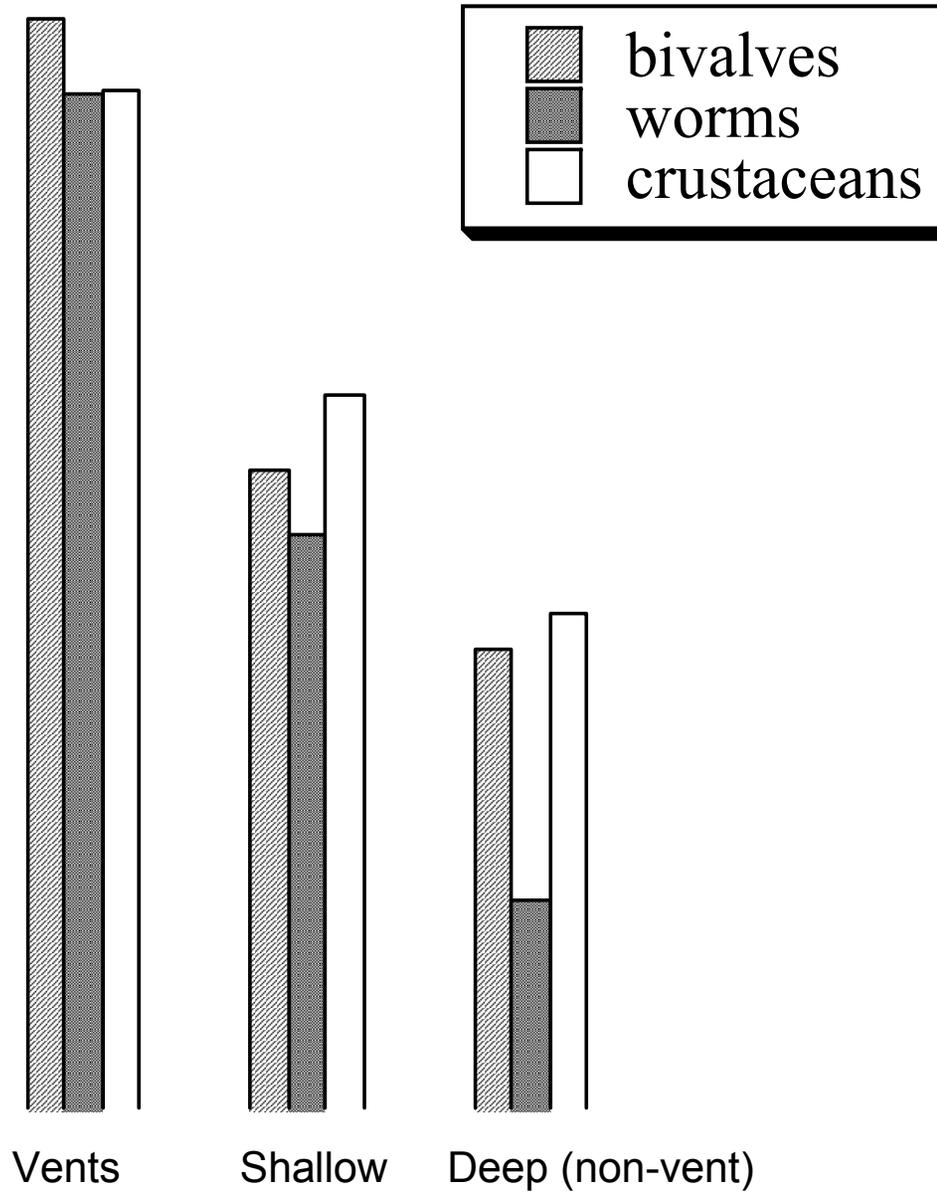
2. Ion transport



# Typical timecourse for acid-base compensation in the blood



Log  
Carbonic  
anhydrase  
activity



1989

QuickTime™ and a  
decompressor  
are needed to see this picture.

**Pörtner, H. O. et al. Respiration Physiology 1990. 81(2):255-73.** Determination of intracellular pH and PCO<sub>2</sub> after metabolic inhibition by fluoride and nitrilotriacetic acid.

**Pörtner, H. O. Respiration Physiology 1990. 81(2) 275-288.** Determination of intracellular buffer values after metabolic inhibition by fluoride and nitrilotriacetic acid.

**Baker et al. 2009. J. Fish Biol.** A validation of intracellular pH measurements in fish exposed to hypercarbia: the effect of duration of tissue storage and efficacy of the metabolic inhibitor tissue homogenate method.

QuickTime™ and a  
decompressor  
are needed to see this picture.

QuickTime™ and a  
decompressor  
are needed to see this picture.

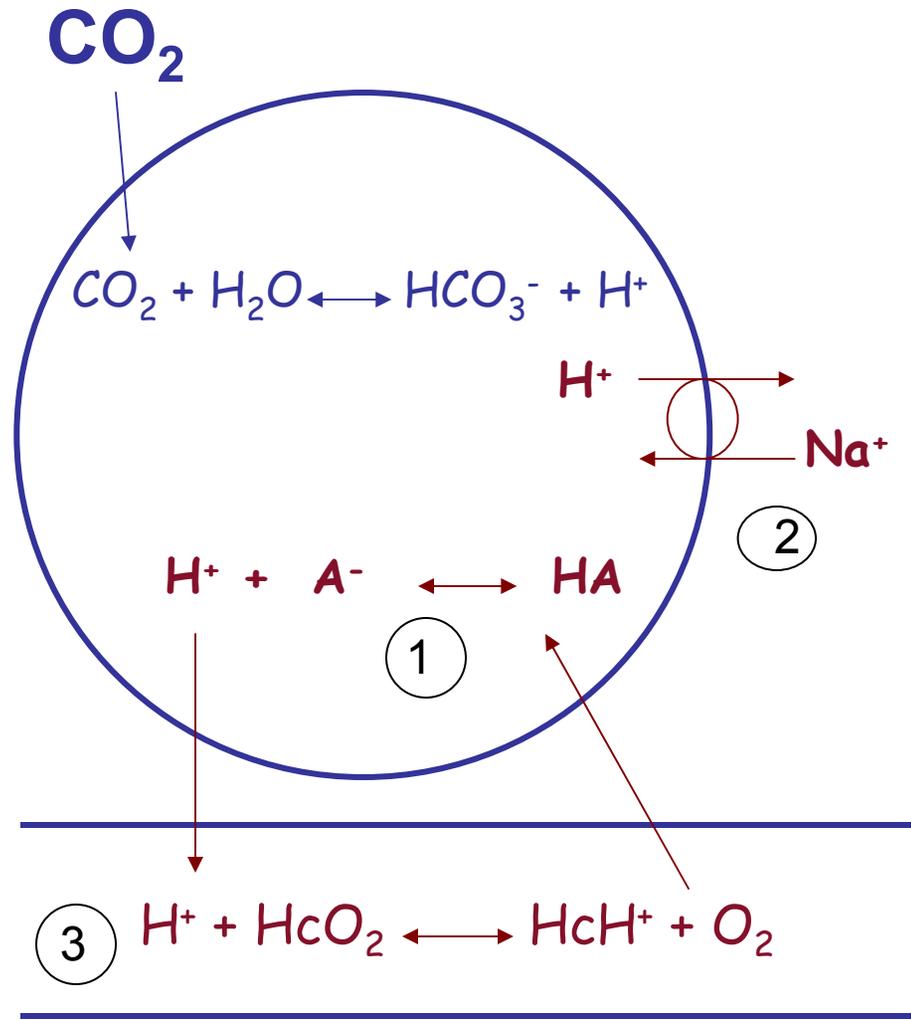
QuickTime™ and a  
decompressor  
are needed to see this picture.

# Combating Acidosis

1. Buffering

2. Proton transport

3. Blood-oxygen binding



# Is a (partially) compensated acidosis good enough?

Time-scale dependent?

Trade-off between ionic- and acid-base balance?

Is there an energetic cost?

Capacity for acclimation?

can organisms produce new isozymes or higher concentrations of relevant enzymes?

# What happens when acid-base regulation fails?

-Enzyme-mediated processes have an evolved pH optimum

QuickTime™ and a  
decompressor  
are needed to see this picture.

-May enhance enzymes quantitatively or qualitatively

# What about at a whole-organism level?

Energetics (Complex, interconnected processes)

QuickTime™ and a  
decompressor  
are needed to see this picture.

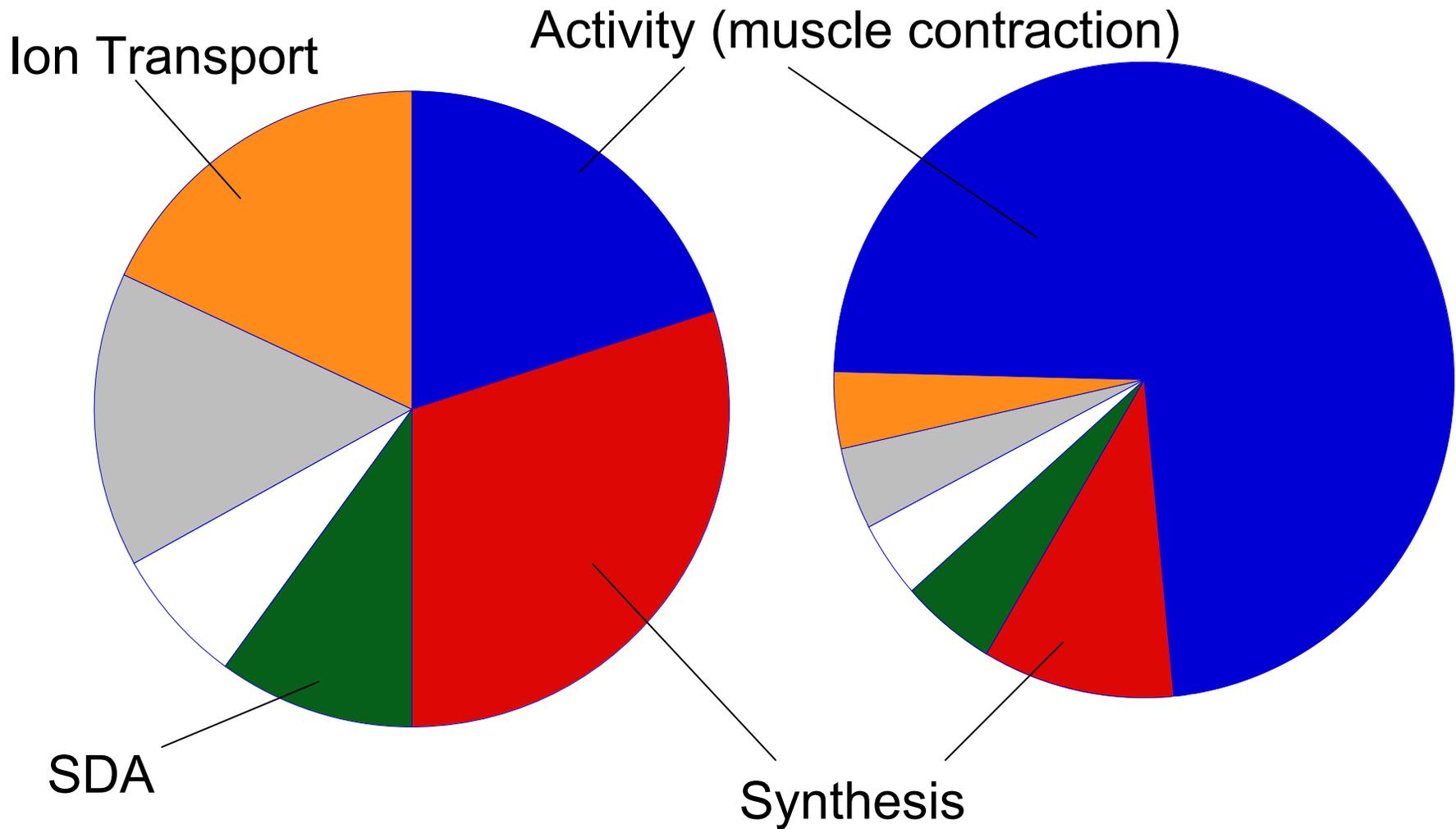
# Metabolic rate as a *cost* to organisms

QuickTime™ and a  
Photo - JPEG decompressor  
are needed to see this picture.

*"Evolution tends to maximize  
metabolic rate, because metabolism  
produces the energy required  
to sustain and reproduce life"*

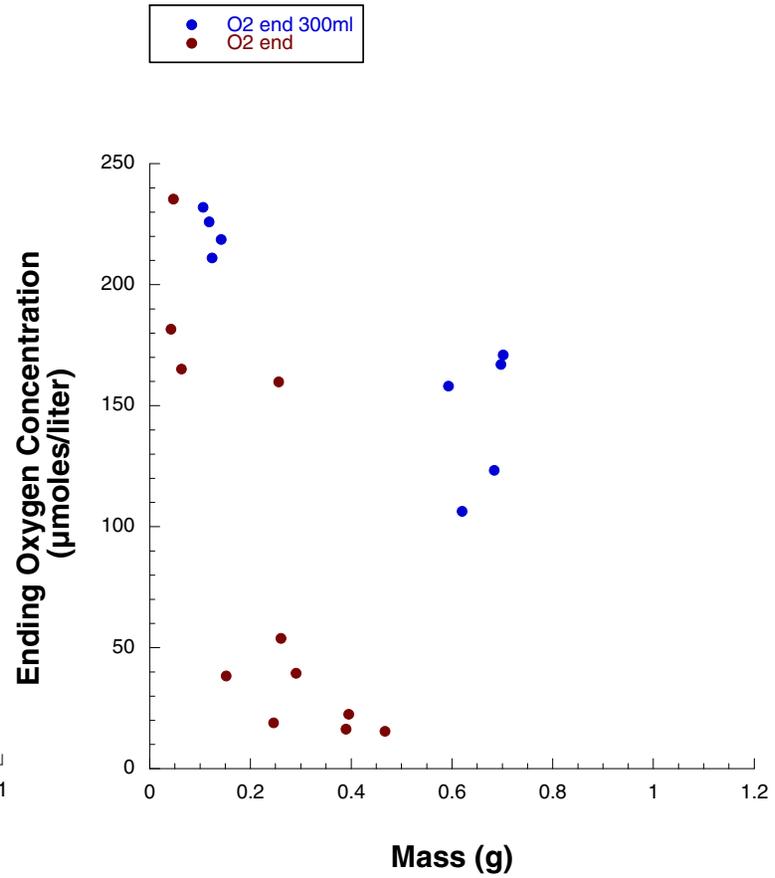
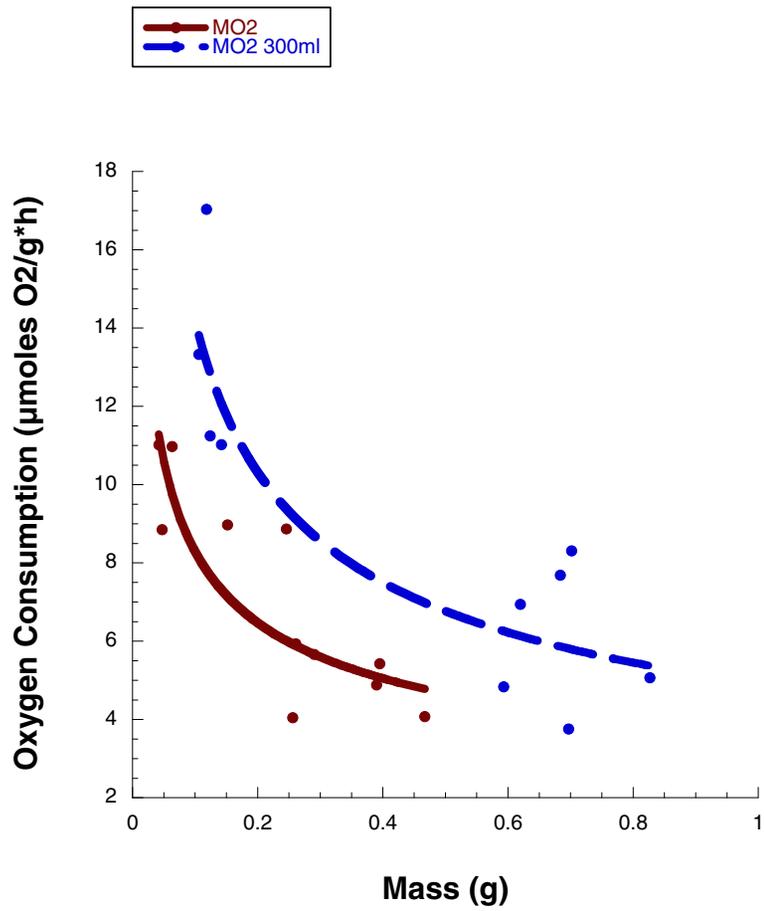
*West et al., 1999, Science*

# Metabolic rate: sum of all costs



# How Do You Measure Metabolic Rate?

1. Duration of experiment (>12 hours)
2. Maintain oxygen well above critical  $PO_2$  (species-specific)
  - $P_{crit}$  typically > 50% saturation
  - Respiratory quotient ~0.7 to 1.0 (CO<sub>2</sub>:O<sub>2</sub>)
3. Temperature within habitat range ( $Q_{10} = 2-3$ )
4. Control activity level (Aerobic scope ~ 5-10X)
  - Basal, Routine, Active, Maximal MR
5. Control feeding (SDA ~ 2-5X BMR)
6. Chamber volume and flow are critical



# Measuring Metabolic Rates

1. Oxygen electrodes vs Optodes
  - expense
  - simplicity
  - stirring
2. Calibration
3. Flow through, static, end-point, or intermittent flow

Closed (static)

Flow-through

QuickTime™ and a  
decompressor  
are needed to see this picture.

Intermittent Flow

# Measuring Metabolic Rates:

## A manual for scientists

John R. B. Lighton

QuickTime™ and a  
decompressor  
are needed to see this pictur

*“It is possible to measure metabolic rates without understanding what you are doing. In doing so you may think, or hope, that the data you acquire are accurate. In fact, this approach is pretty much the rule....but my hope is that this text will discourage this approach”*

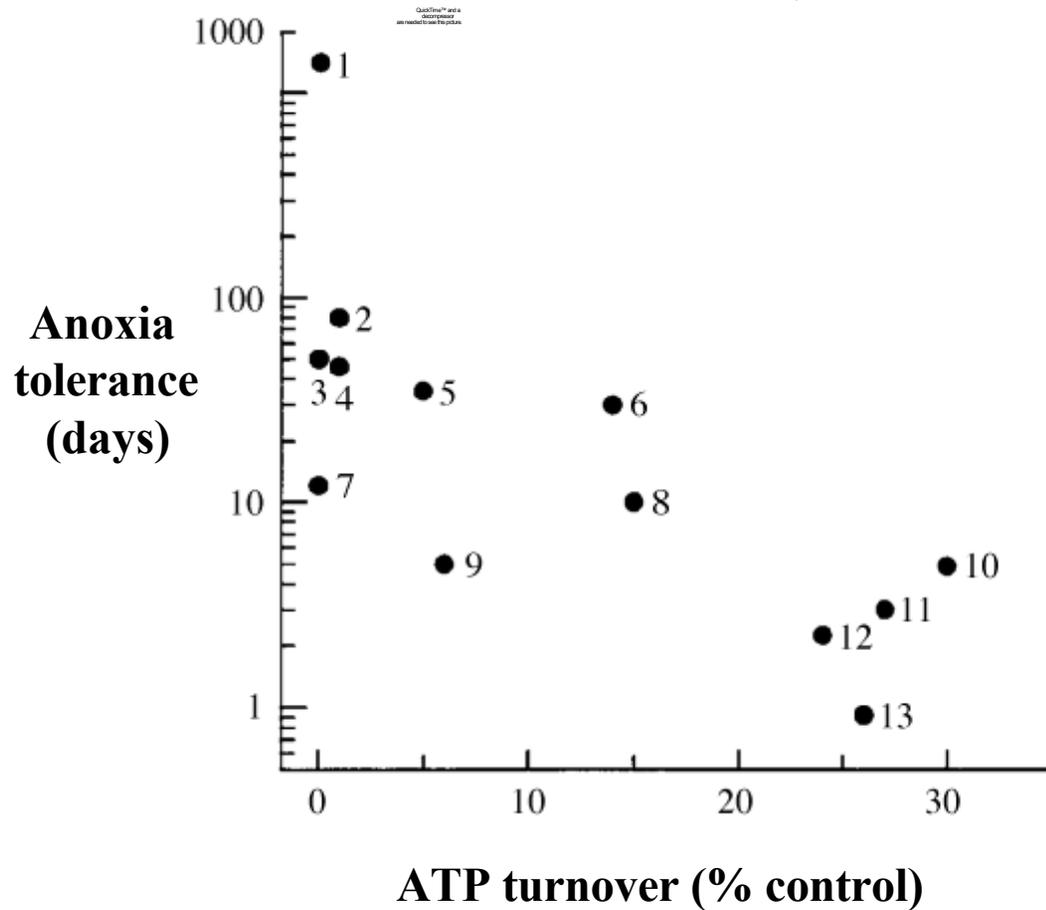
[www.respirometry.org](http://www.respirometry.org)

## Two cases where $\text{CO}_2$ may reduce $\text{MO}_2$ :

1. Metabolic Suppression as an adaptive response to oxygen limitation - often triggered by  $\text{CO}_2$  or pH.
2. Oxygen transport limitation due to pH effect on respiratory protein in blood.

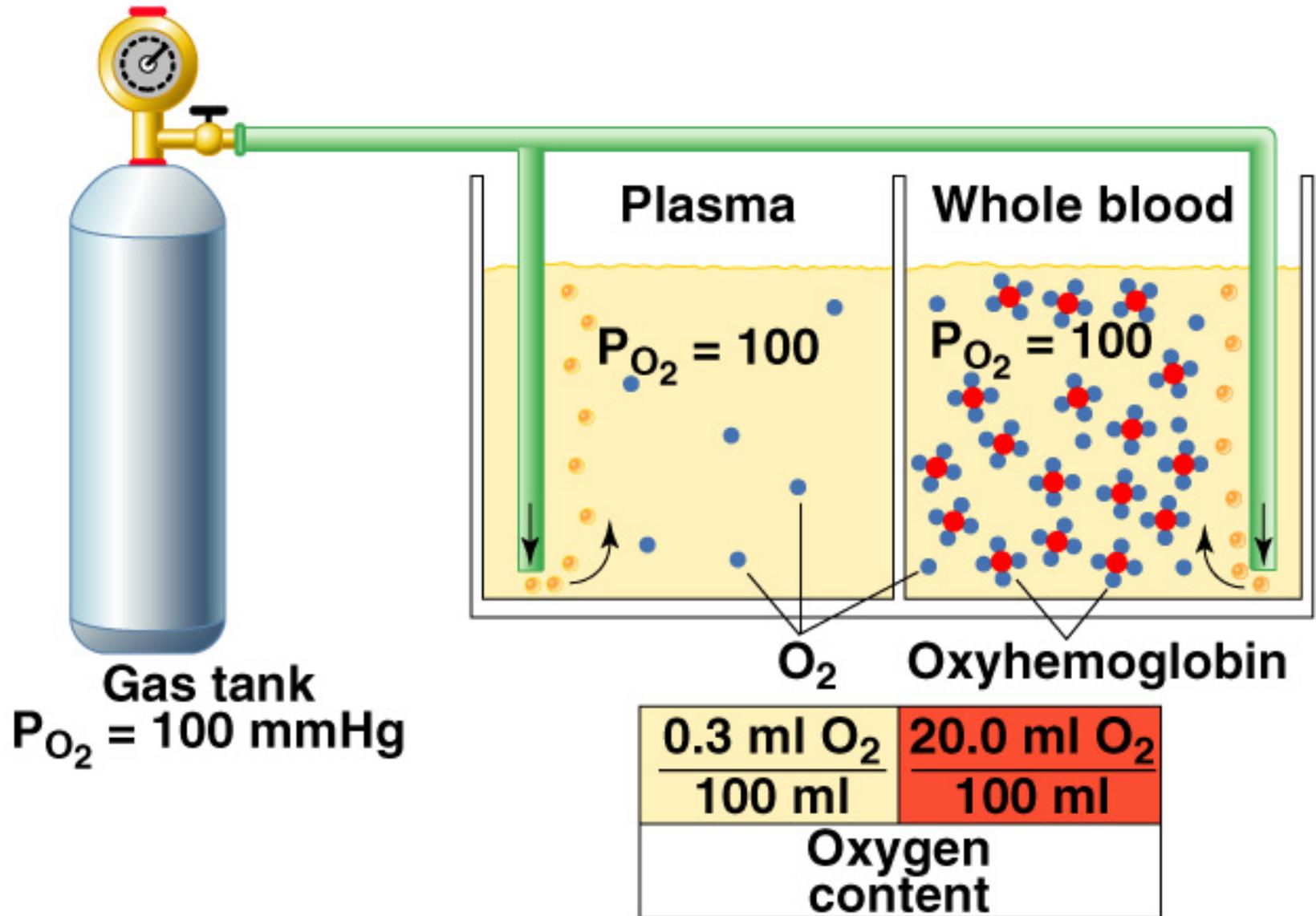
# Metabolic Suppression

Anoxia tolerance correlates with metabolic suppression

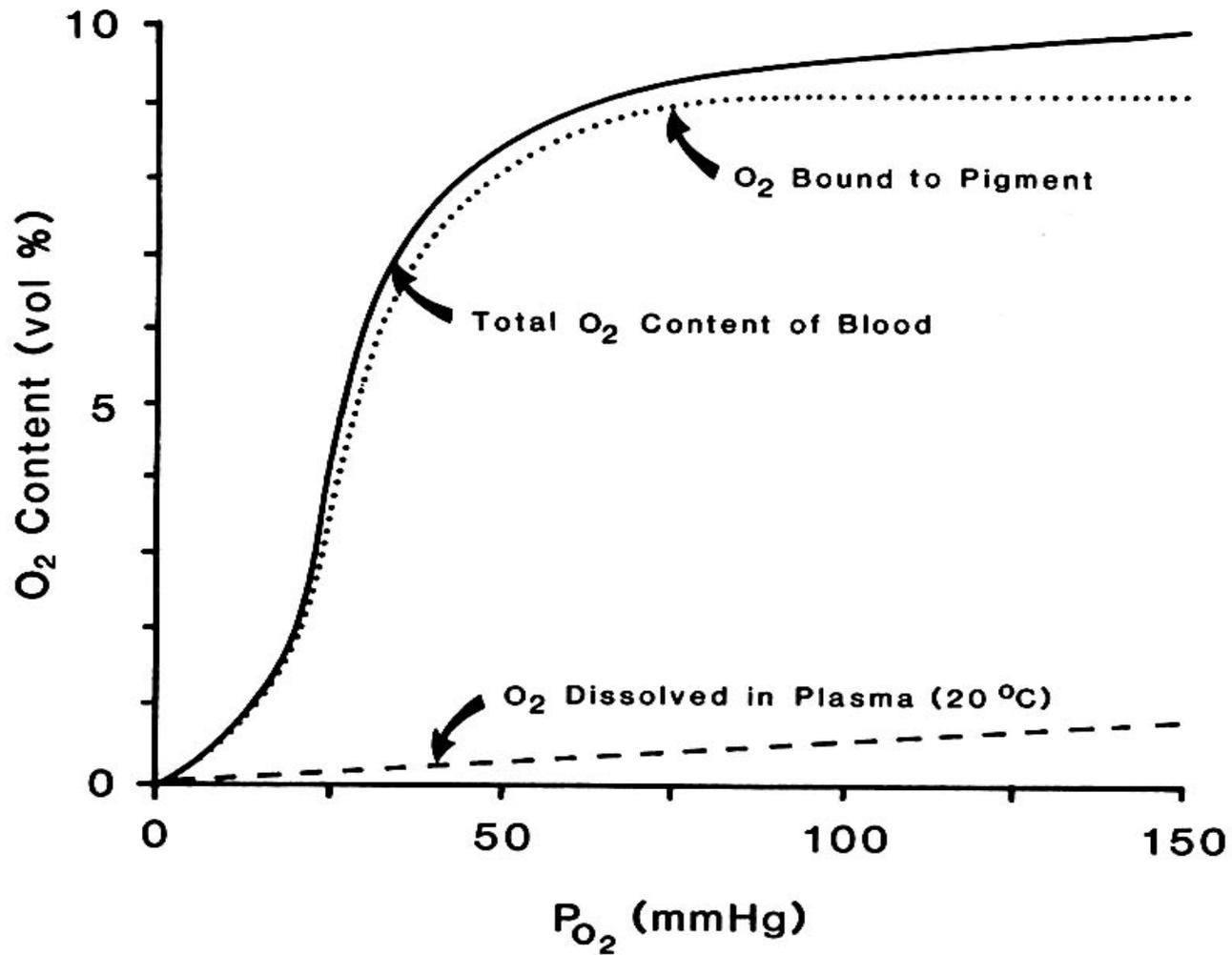


# Oxygen-limitation of metabolism due to pH effect on protein

QuickTime™ and a  
TIFF (Uncompressed) decompressor  
are needed to see this picture.

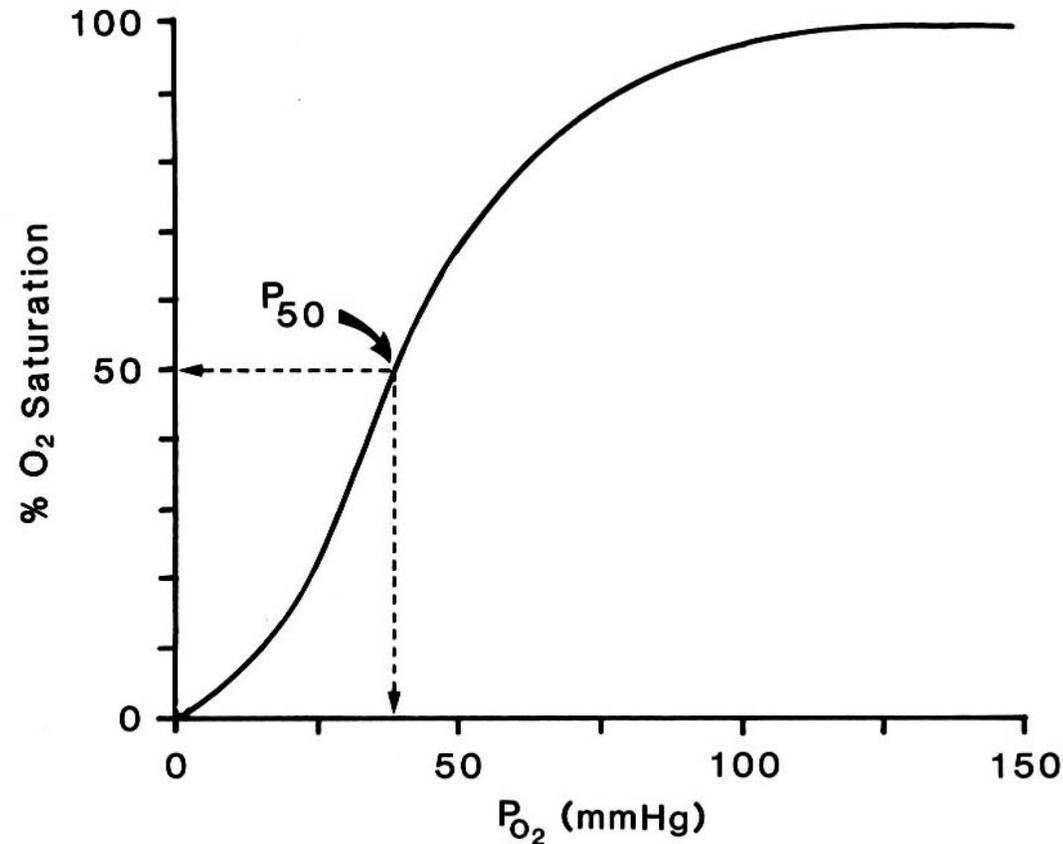


# RESPIRATORY PIGMENT FUNCTION

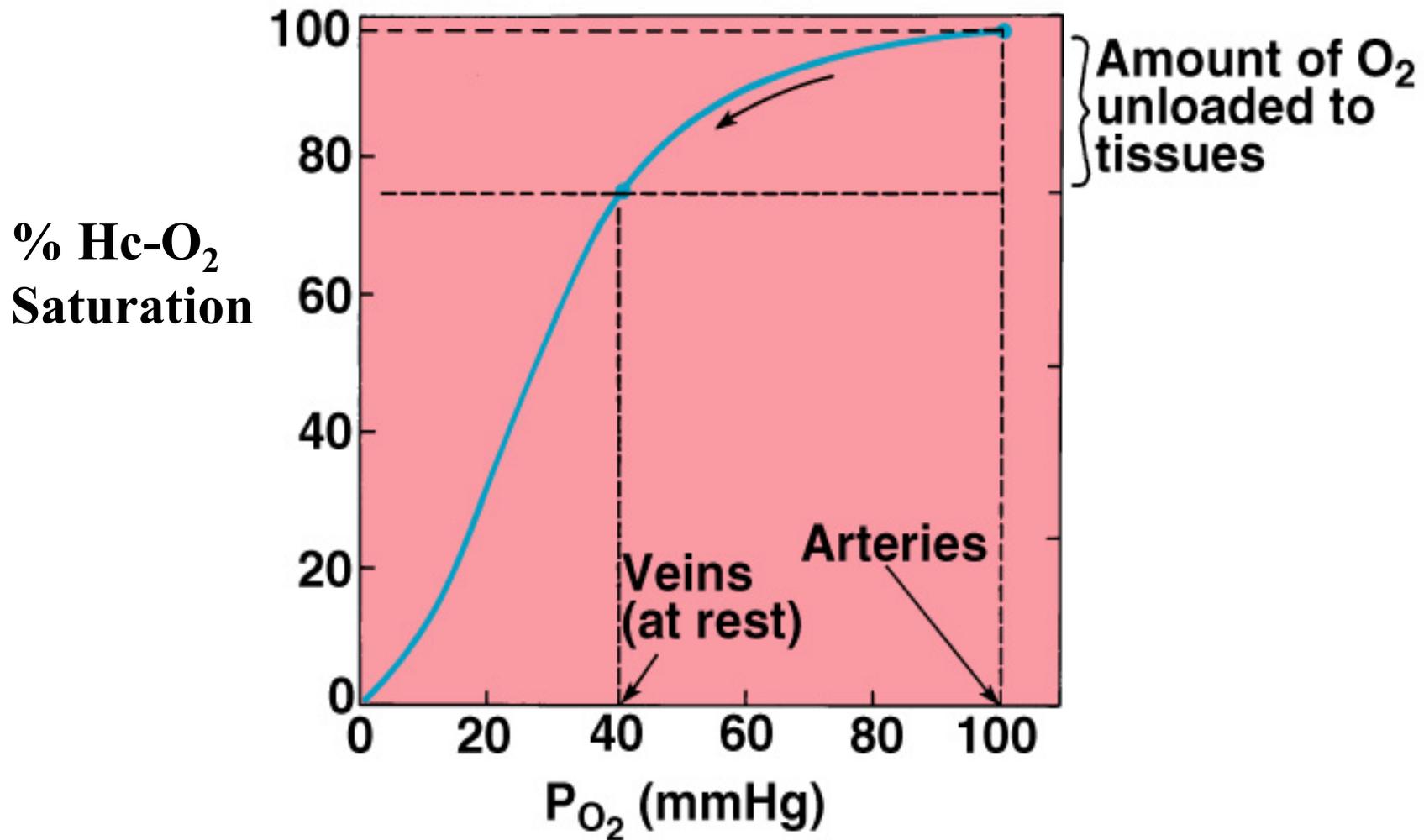


# Oxygen Binding Curves

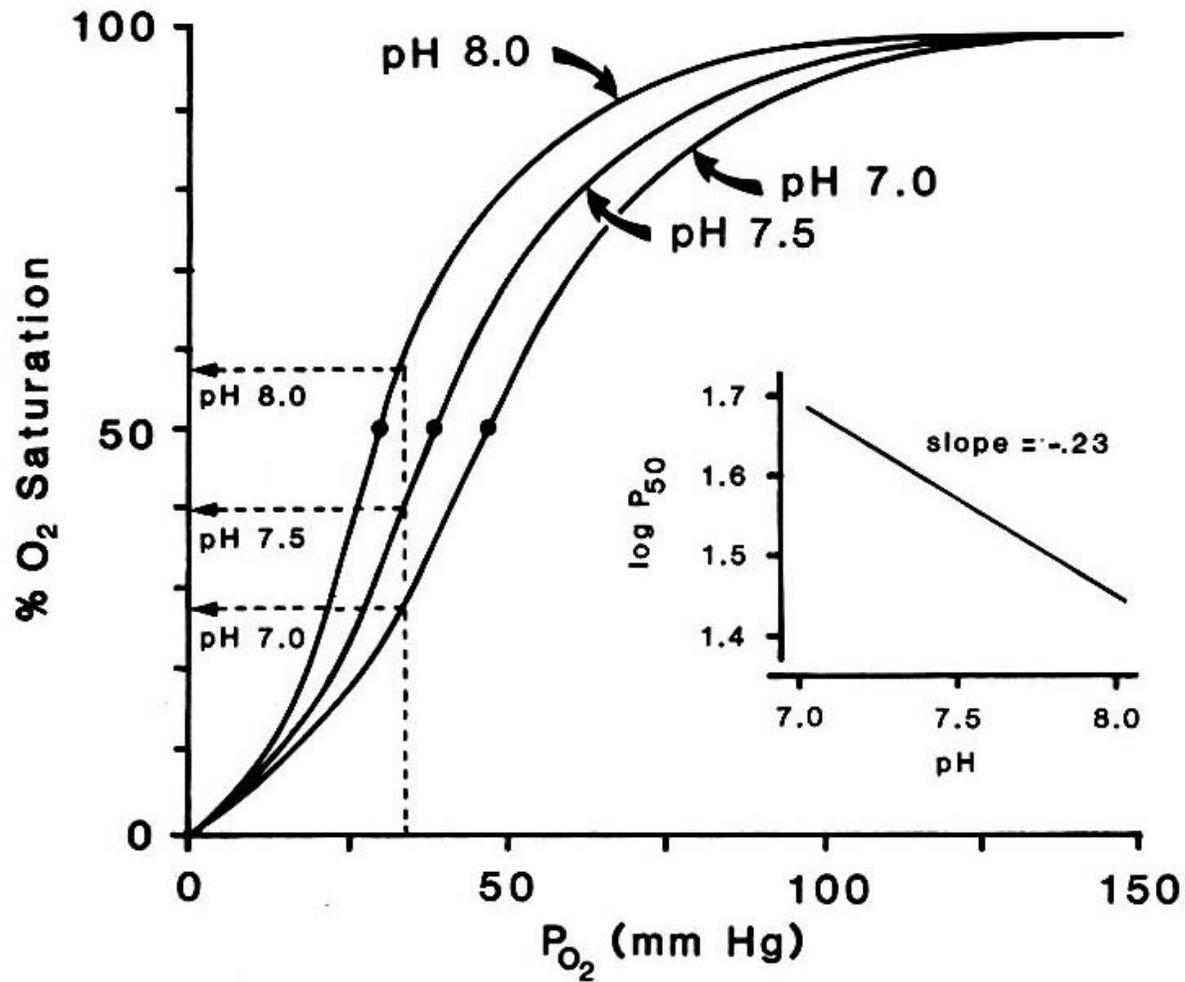
The affinity of an oxygen-binding pigment for  $O_2$  is given as the  $P_{50}$ , the  $PO_2$  required to saturate half the  $O_2$  binding sites.



# Oxygen Binding Curves

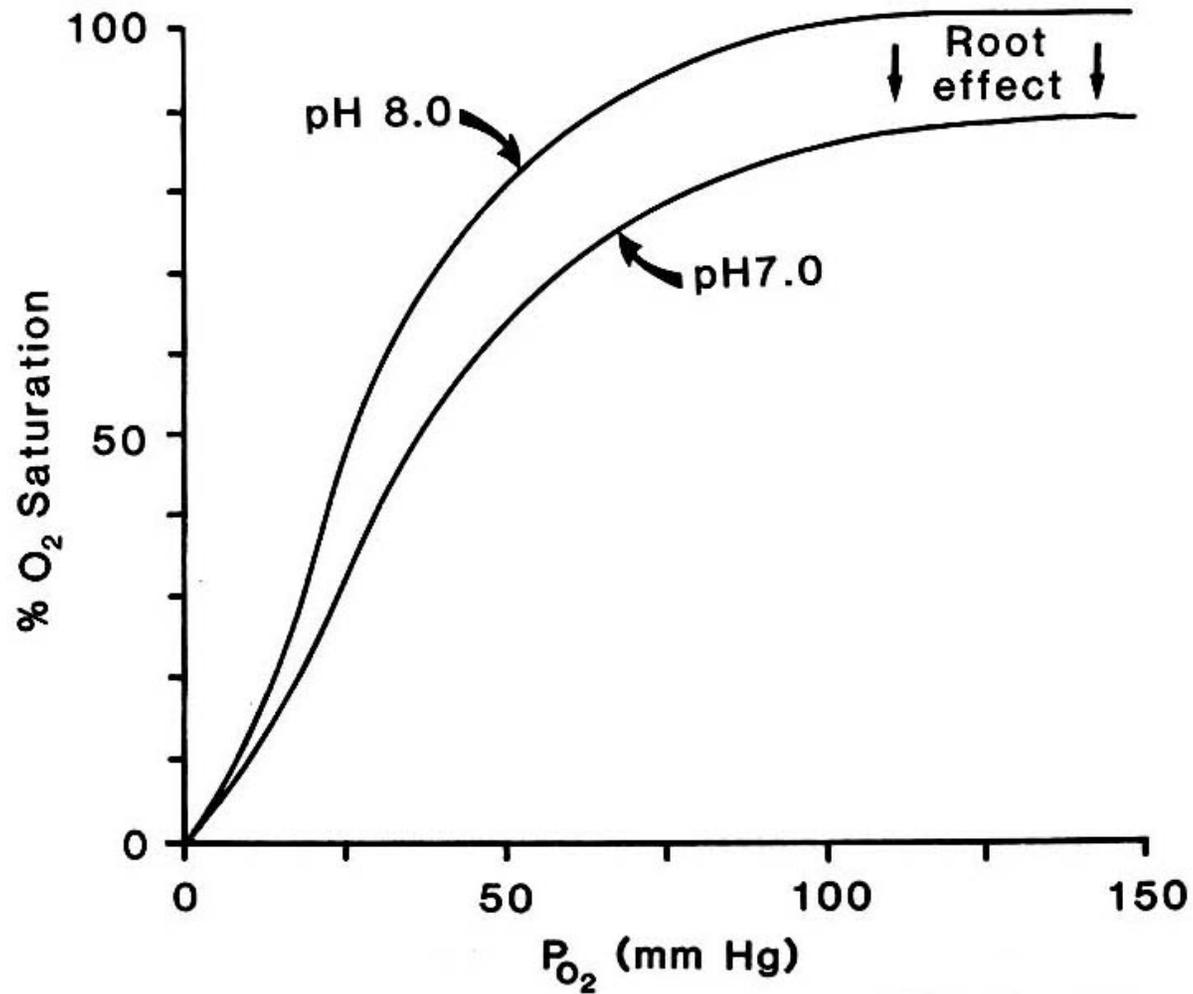


# pH: THE BOHR EFFECT



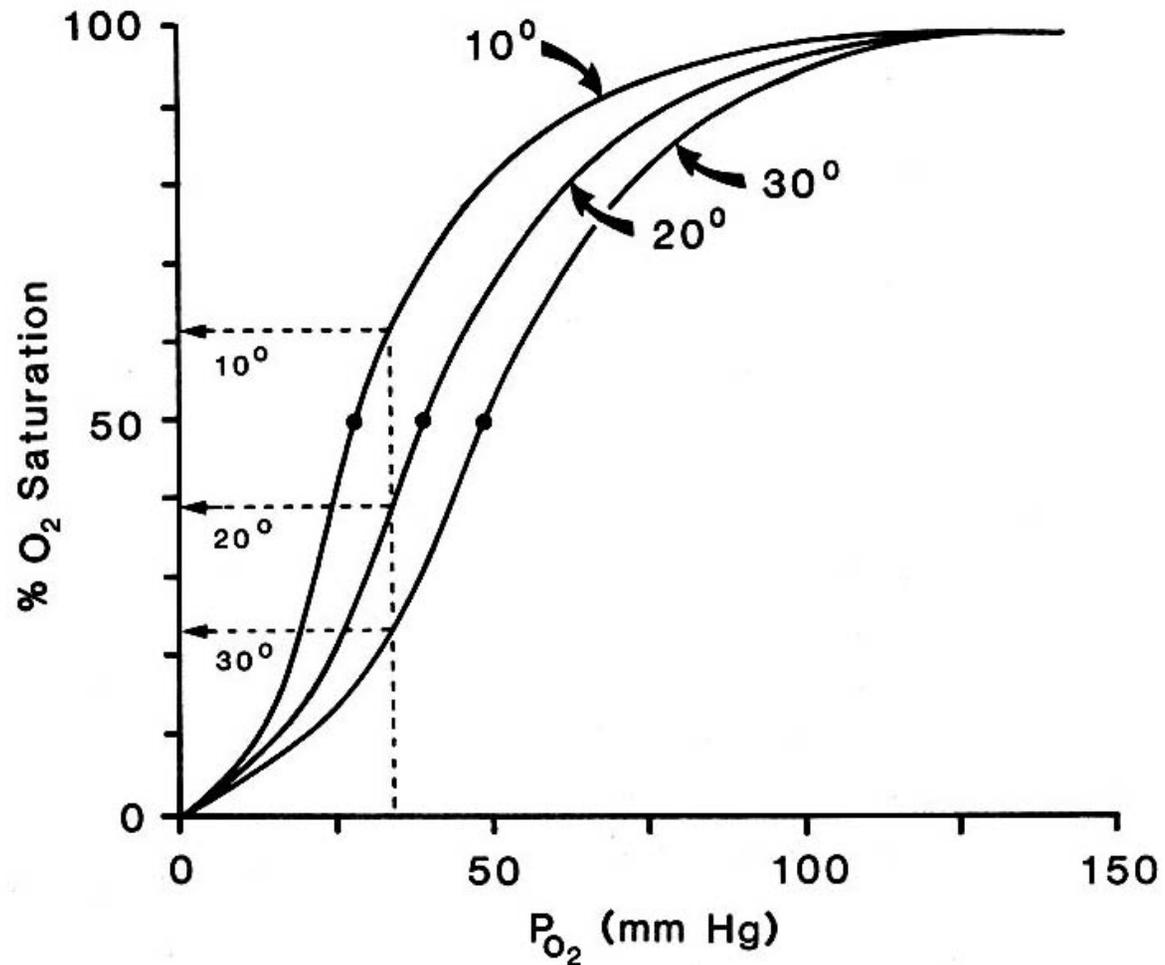
(From Burggren *et al.* 1993)

# pH: THE ROOT EFFECT



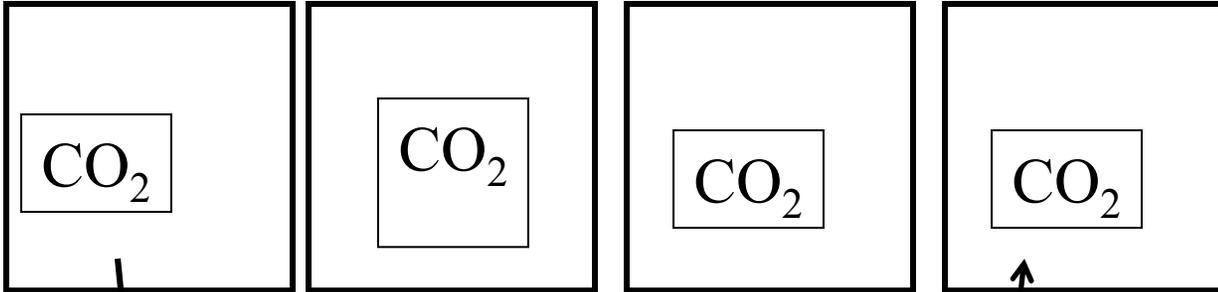
(From Burggren *et al.* 1993)

# TEMPERATURE EFFECTS

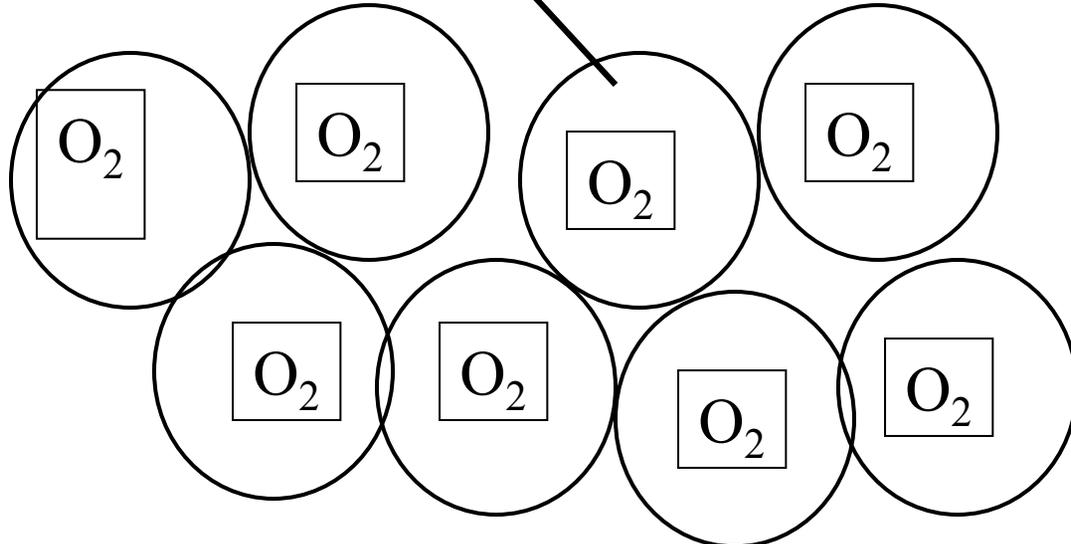
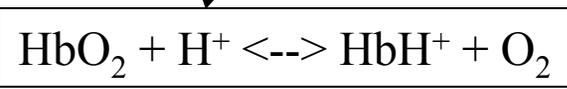


(From Burggren *et al.* 1993)

**Tissues**



High  $\text{CO}_2$   
Low pH  
Low  $\text{O}_2$



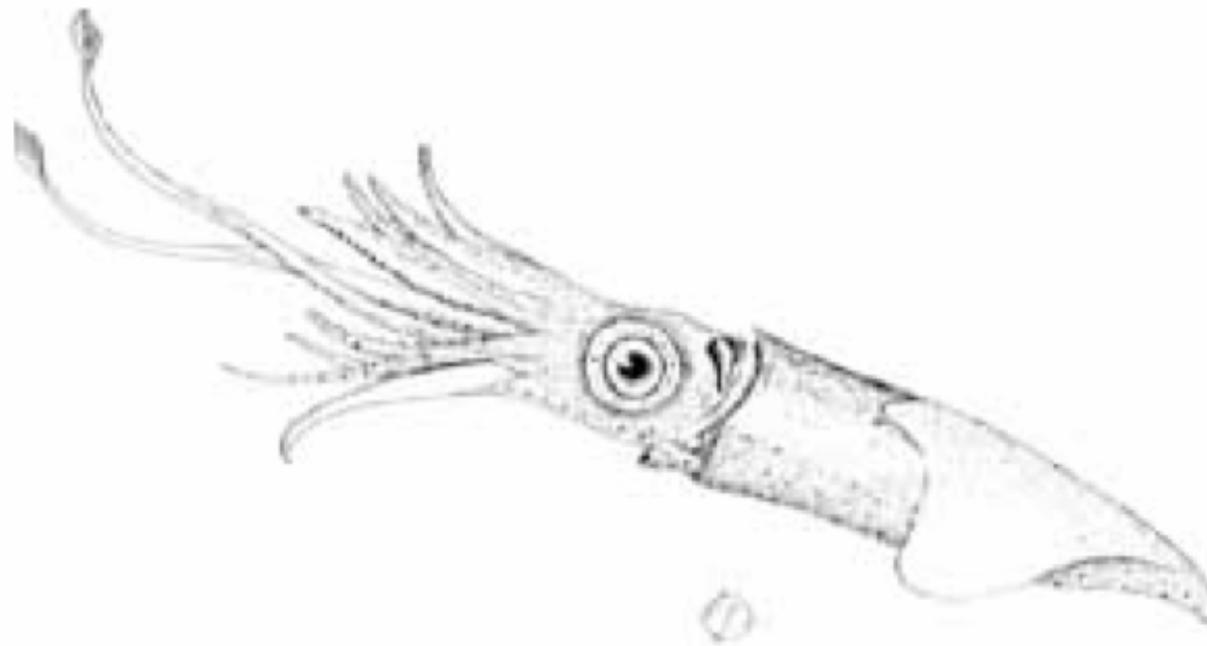
Low  $\text{CO}_2$   
High pH  
High  $\text{O}_2$

**Gills**

# Squids as extreme animal models

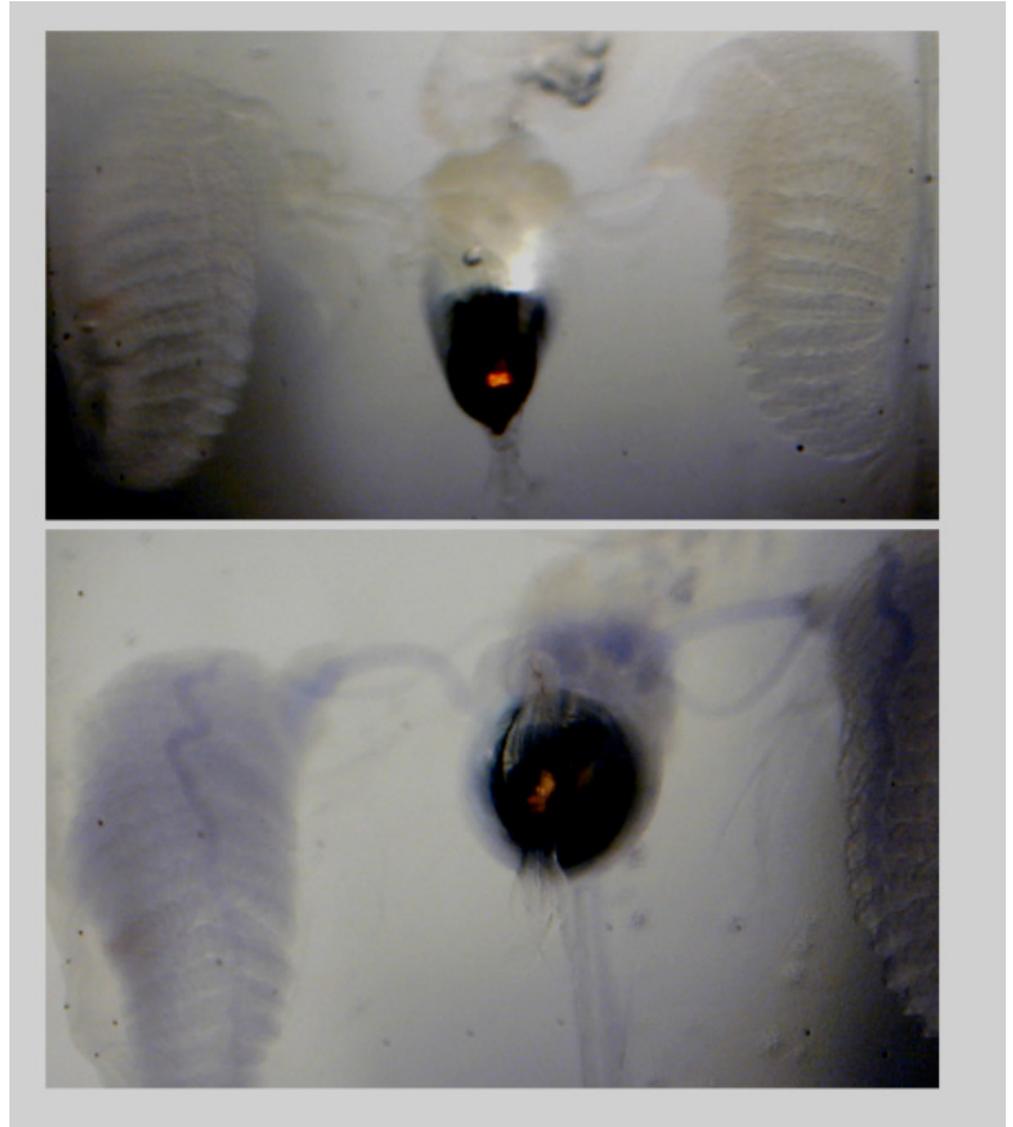
***“The fine-tuning of hemocyanin function underlines the dependence of squid on low environmental CO<sub>2</sub>”***

***Pörtner and Reipschläger, 1997***



# Squids as extreme animal models: "The edge of oxygen limitation"

- High O<sub>2</sub> Demand
- Low carrying capacity
- No venous reserve



# Squids vs Fishes: Similar thrust, less efficient

QuickTime™ and a  
TIFF (Uncompressed) decompressor  
are needed to see this picture.

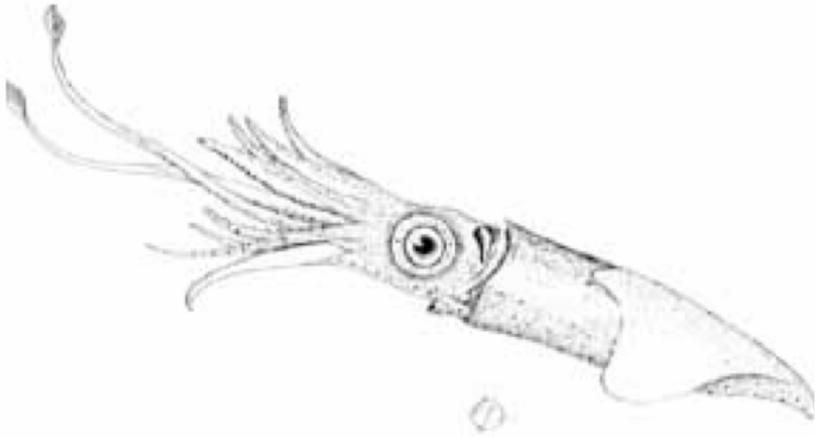


**Fins move  
more water  
slowly**

QuickTime™ and a  
TIFF (Uncompressed) decompressor  
are needed to see this picture.



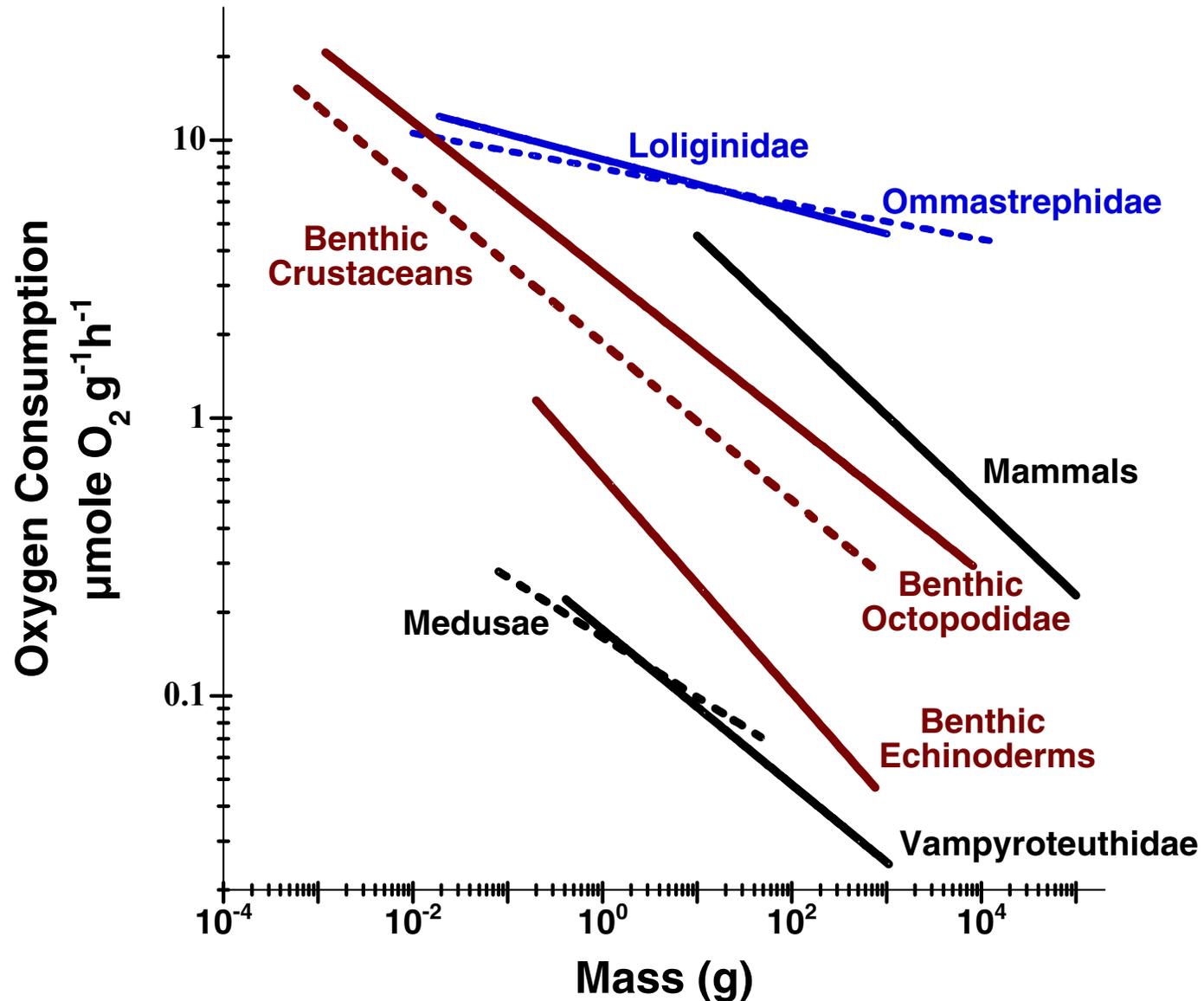
**Jet propulsion  
moves less water  
at greater speeds.**



*Squids must consume twice  
as much oxygen to go half as fast  
as a similar sized fish*

**-O'Dor and Webber, 1986**

# High metabolic rates in squids





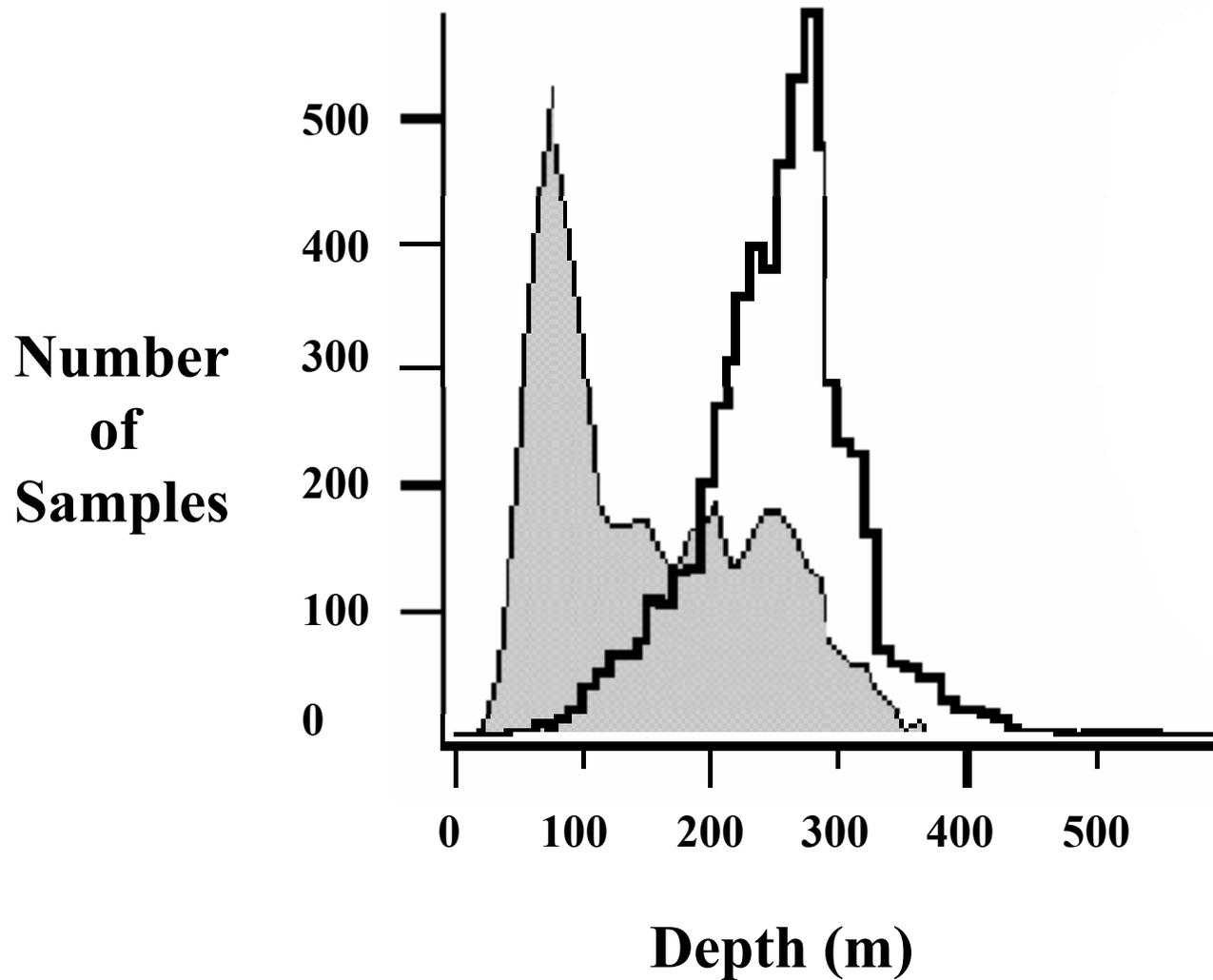
*Squids may operate at environmental limits  
of temperature, oxygen availability  
and body size*

**-Pörtner, 2002**

# *Dosidicus gigas*: Extreme animal in an Extreme Environment

QuickTime™ and a  
H.264 decompressor  
are needed to see this picture.

# Day and Night Depth Distribution of *Dosidicus gigas*



Gilly et al., 2006, MEPS

# Oxygen Minimum also a CO<sub>2</sub> *Maximum* Zone

pCO<sub>2</sub> (ppmv)

QuickTime™ and a decompressor are needed to see this picture.

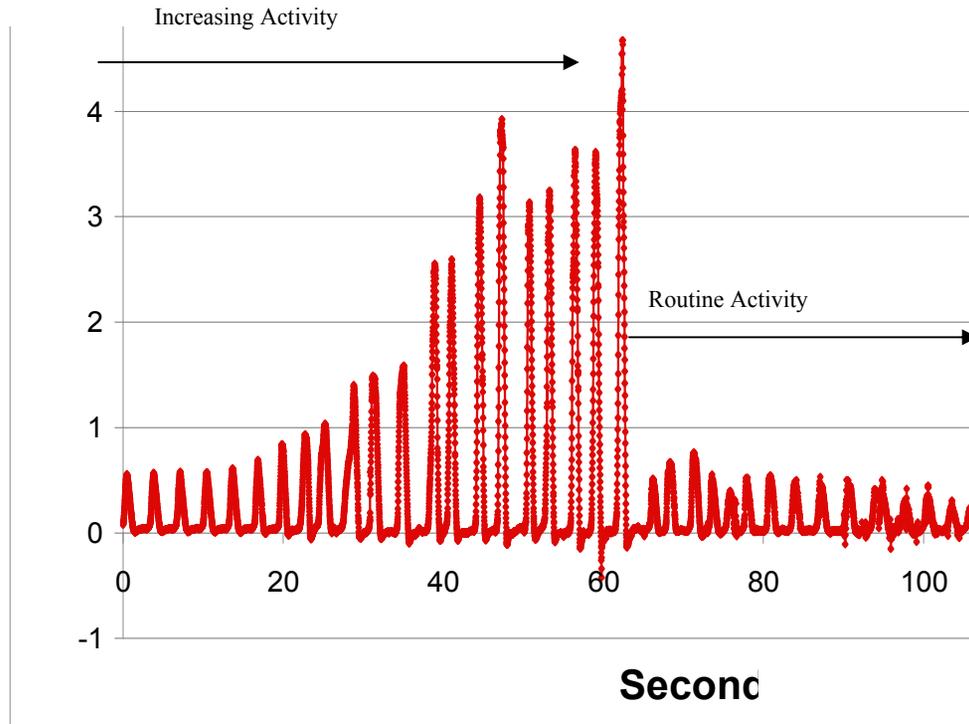
Depth (m)

# HOT TUB

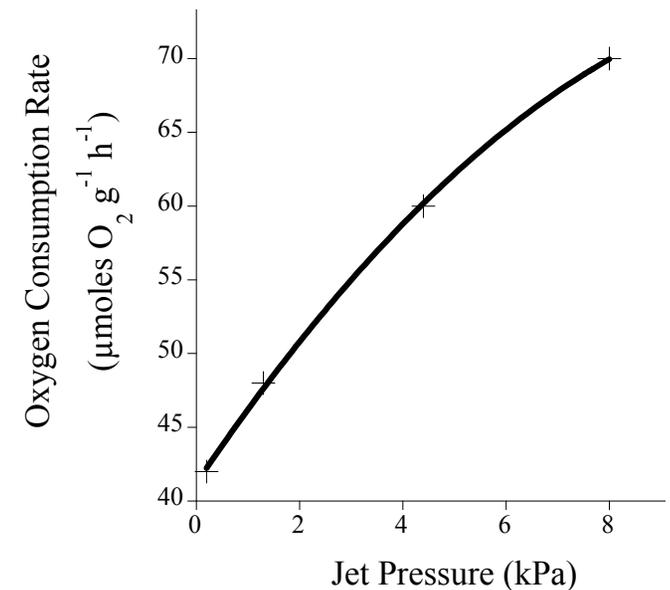
## Humbolt Oxygen Temperature Utilization Basin



# Generating Field Metabolic Rates



The pressure generated by mantle contractions during swimming is correlated with metabolism

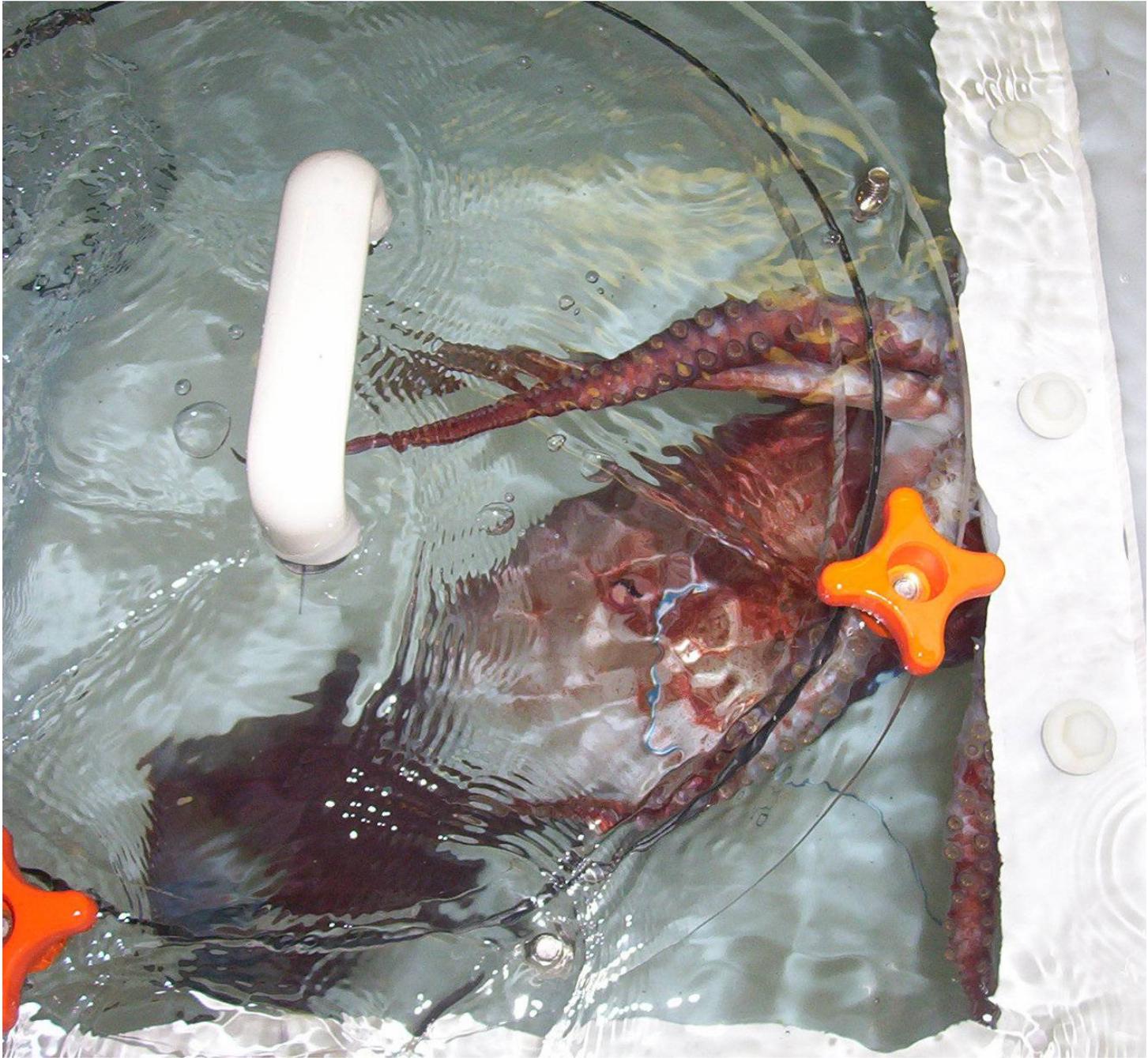




QuickTime™ and a  
YUV420 codec decompressor  
are needed to see this picture.



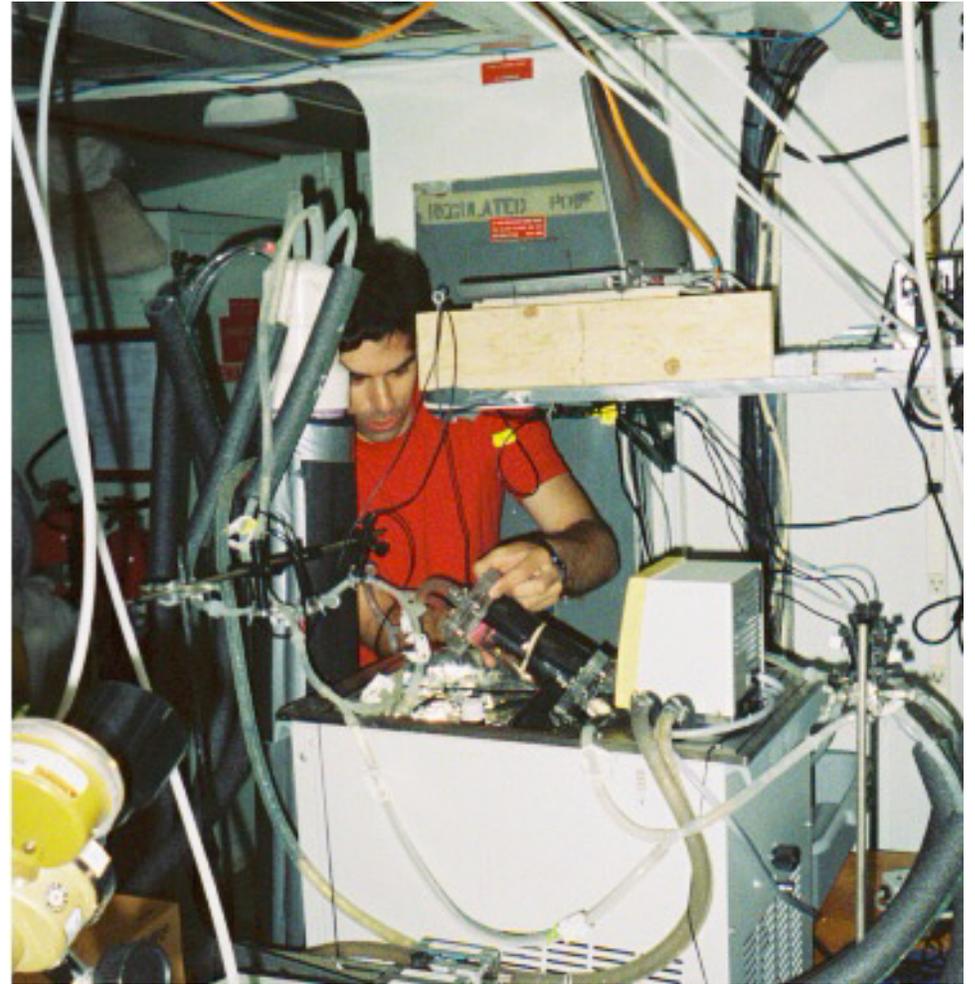
16 11:39 PM

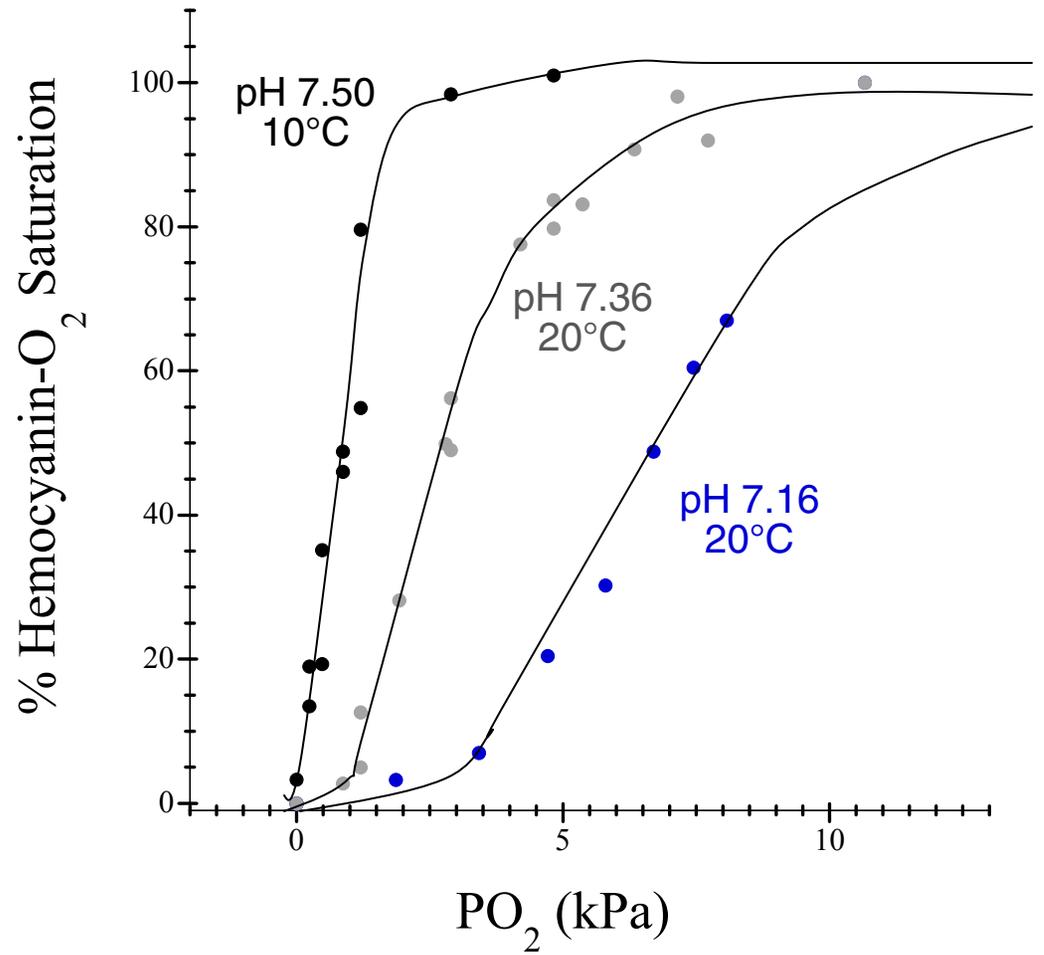
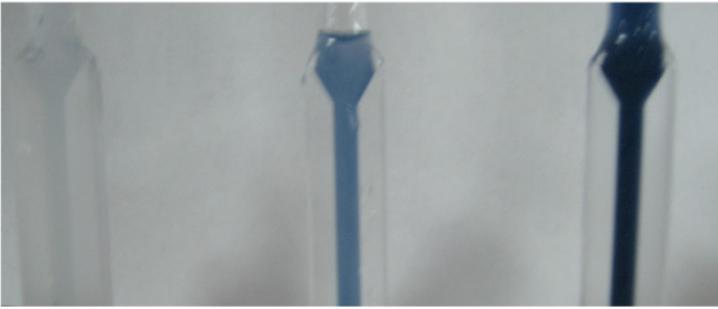




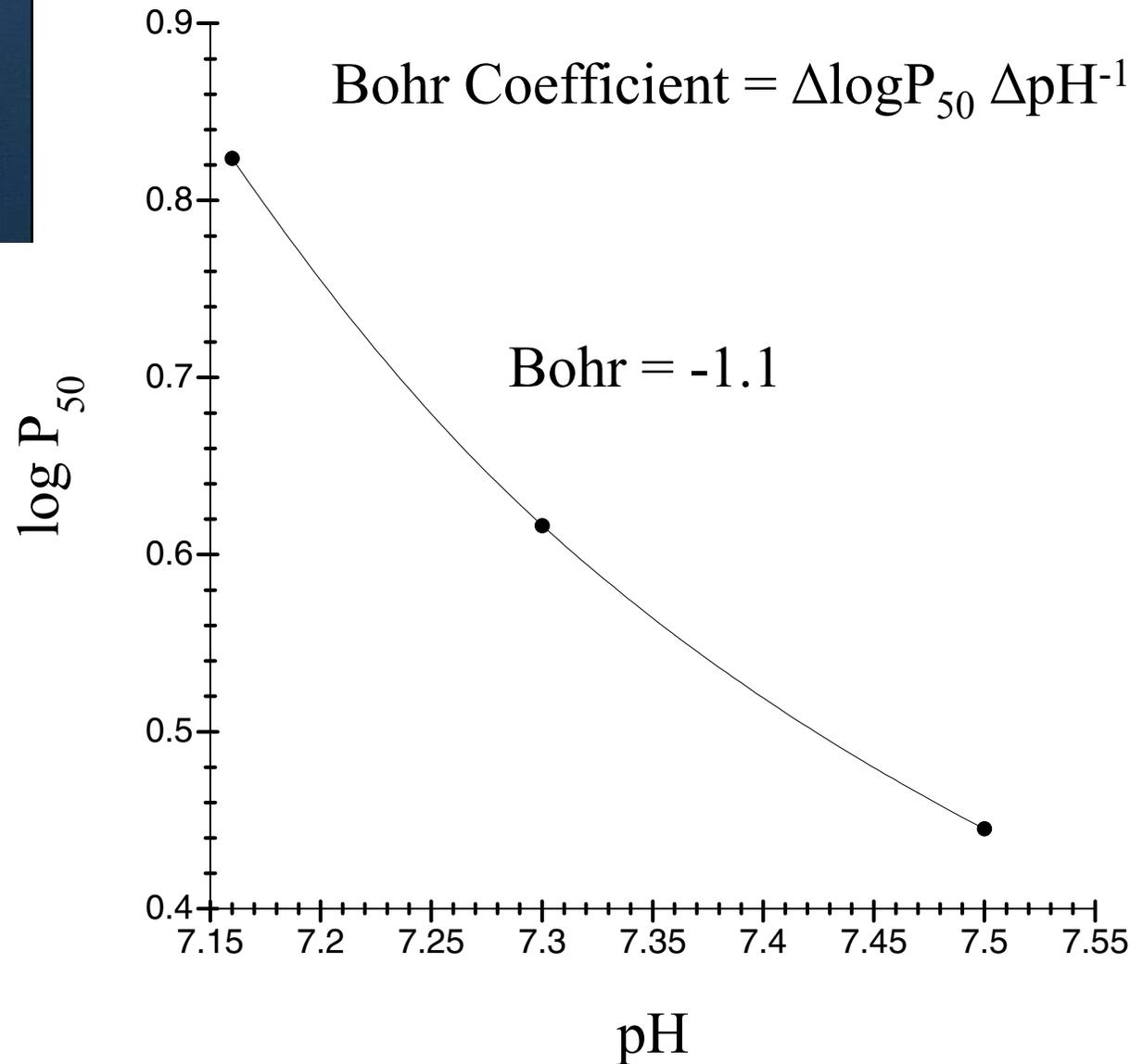
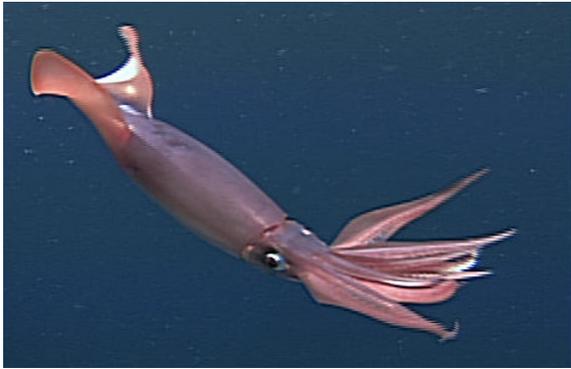
**Dr. Rui Rosa**

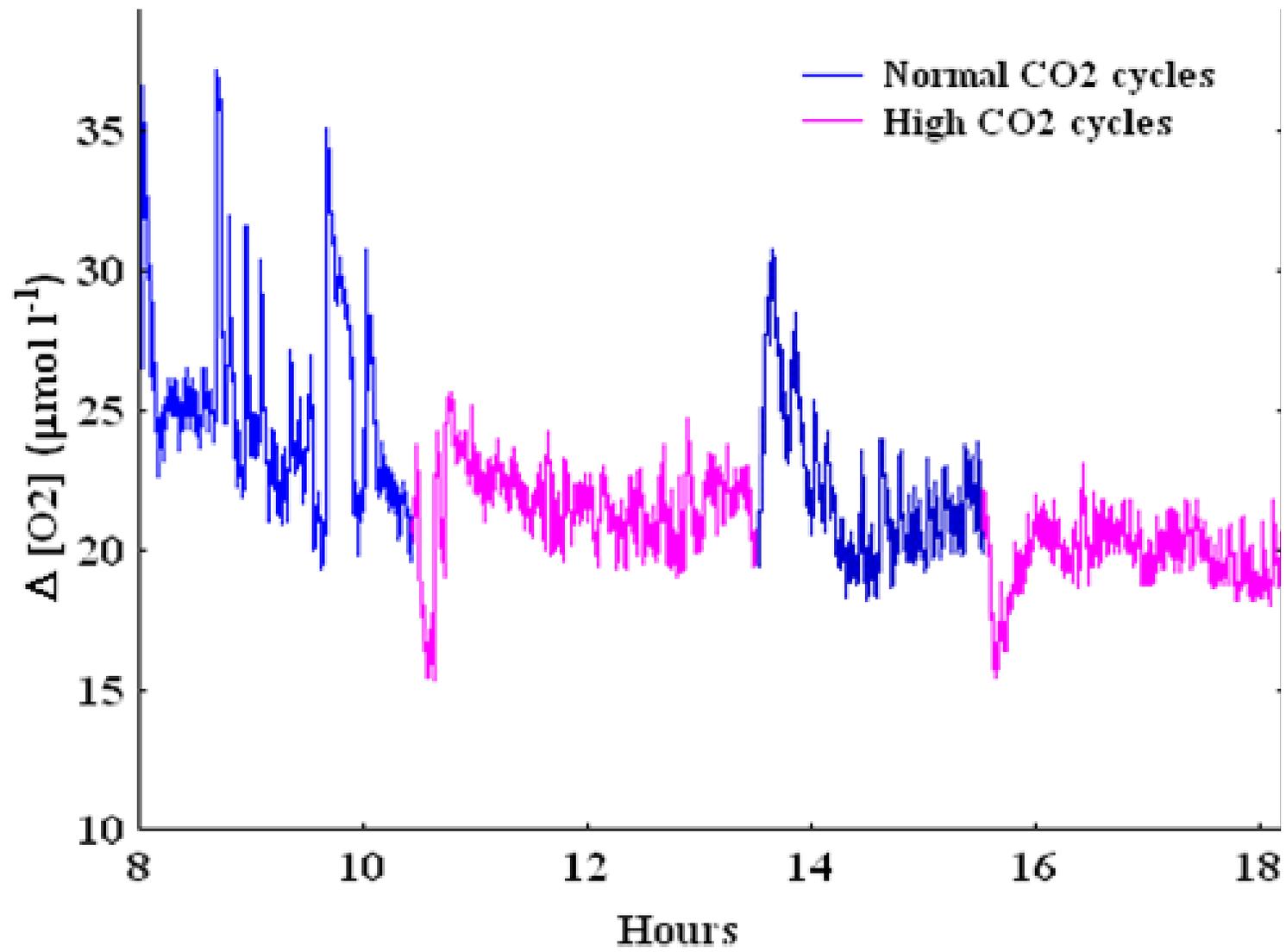
**Metabolic depression  
at high  $CO_2$ ?**

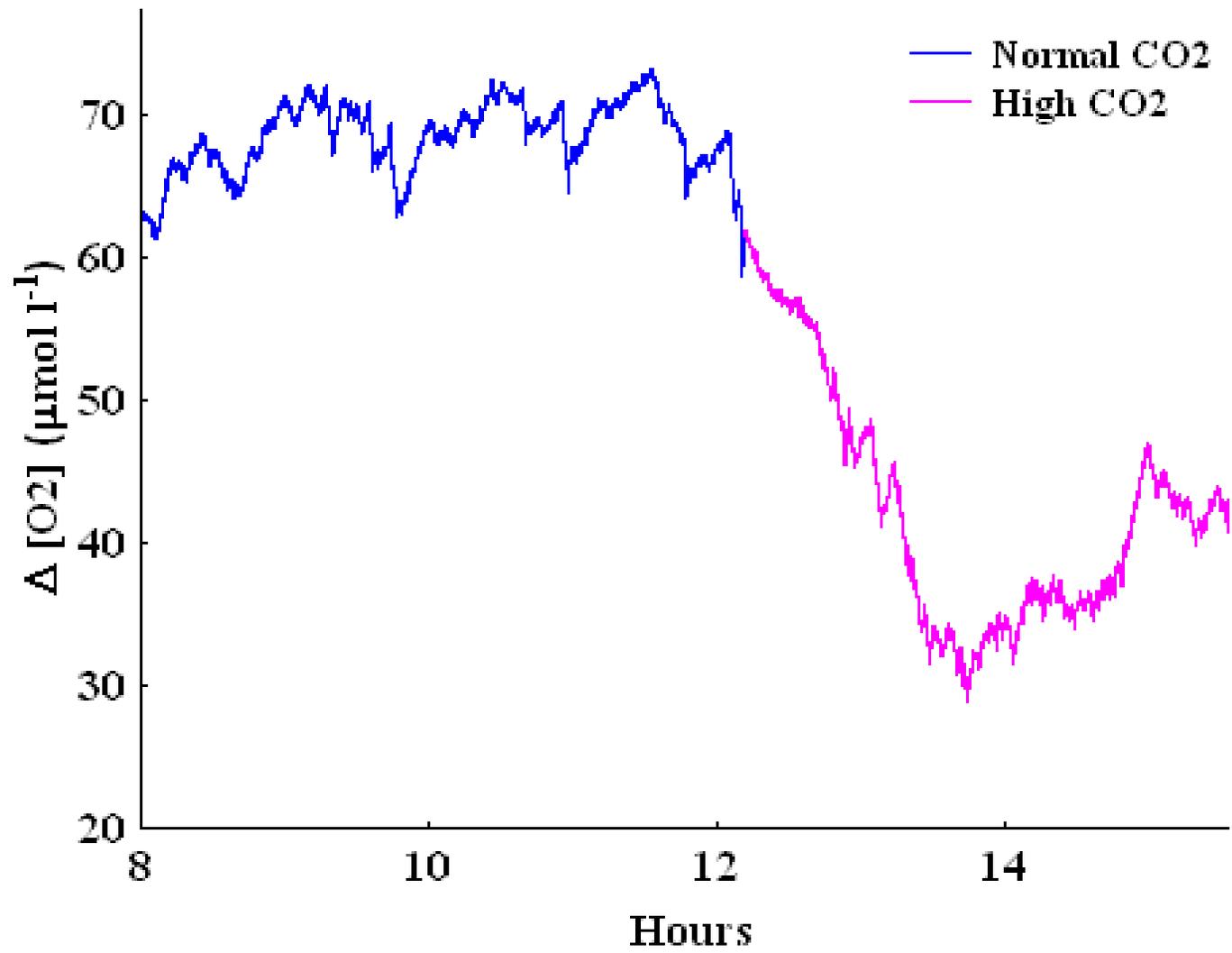




# Highly pH-sensitive oxygen binding

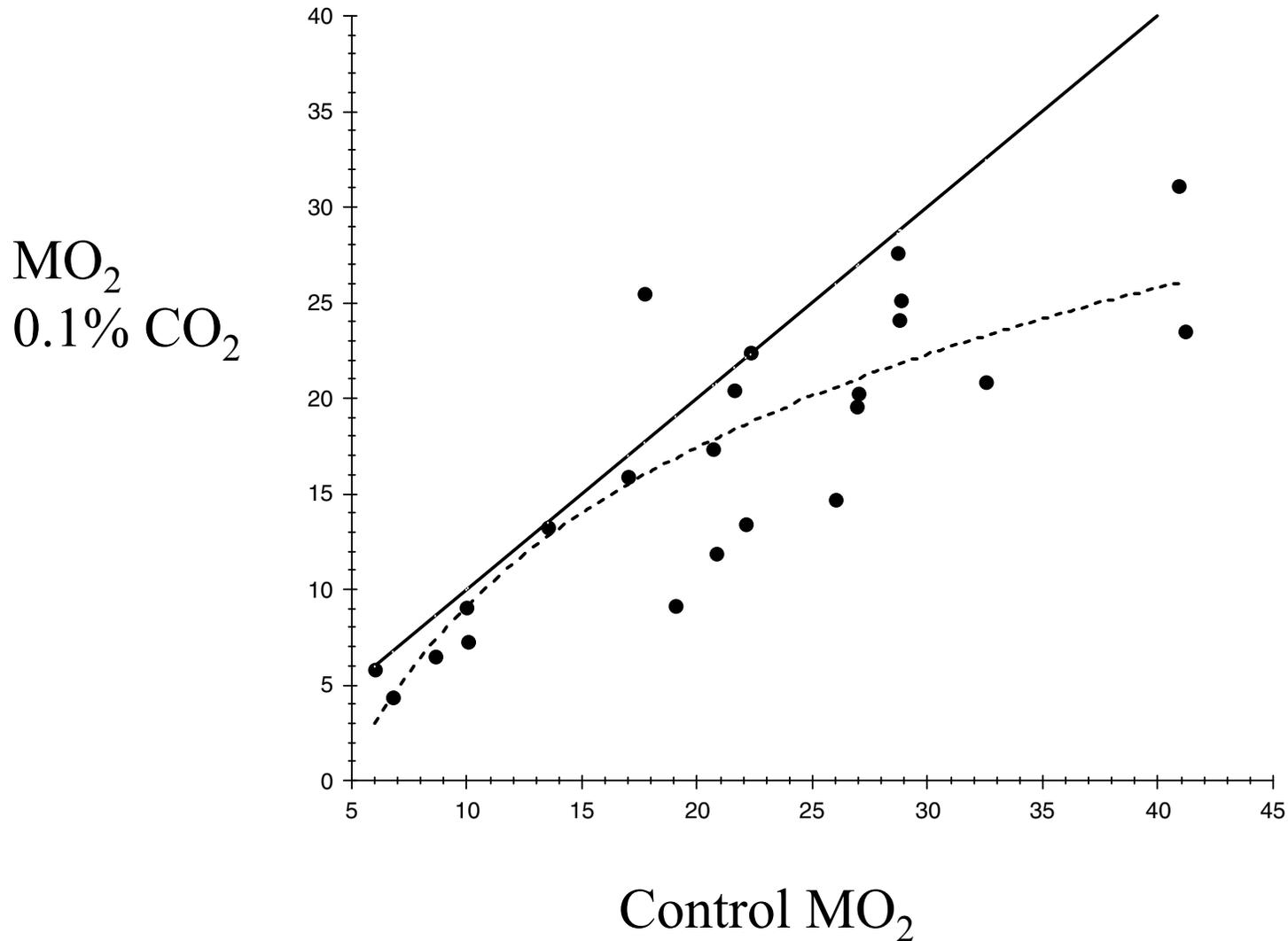




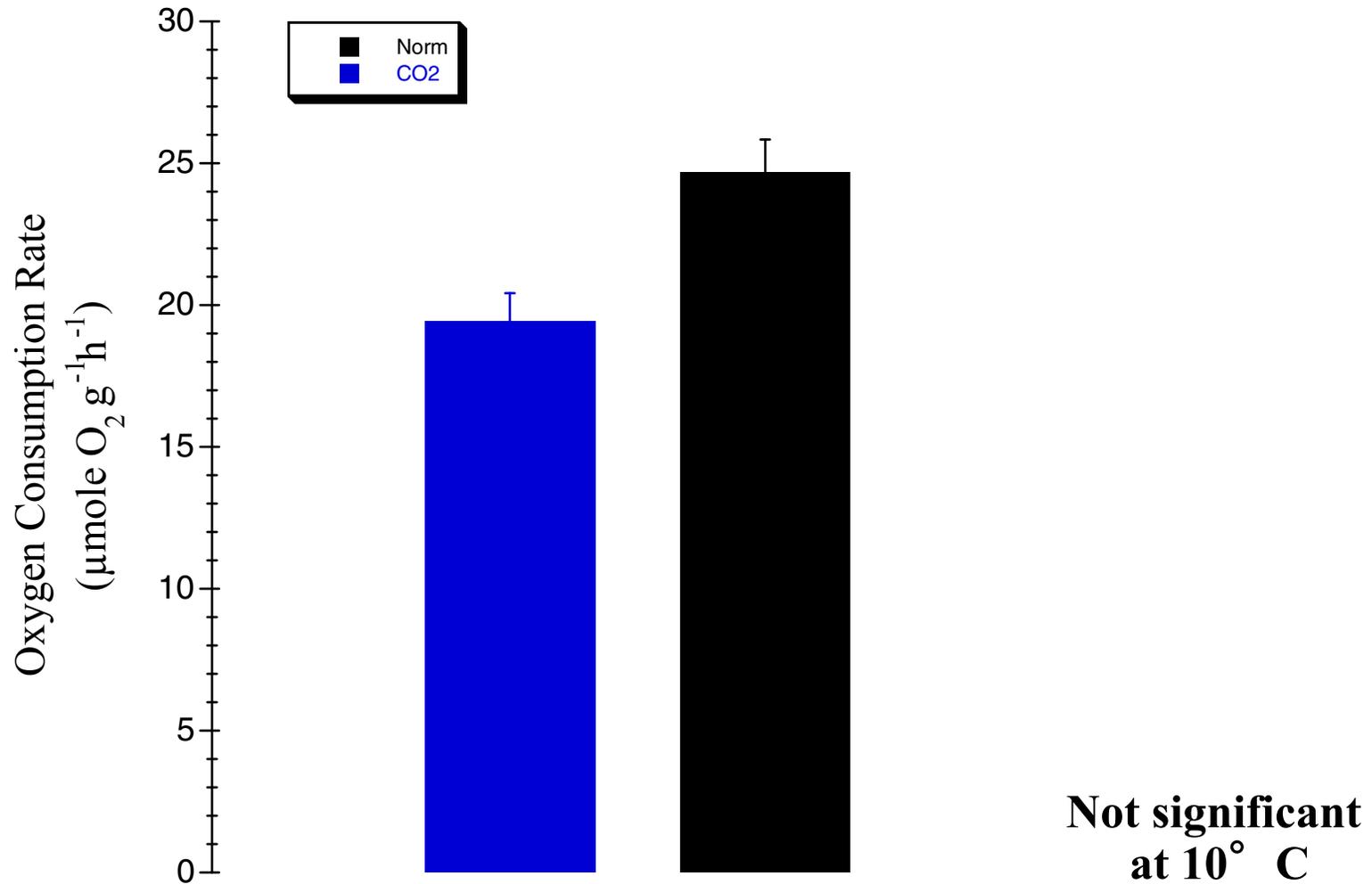


# $CO_2$ effect on oxygen consumption rates

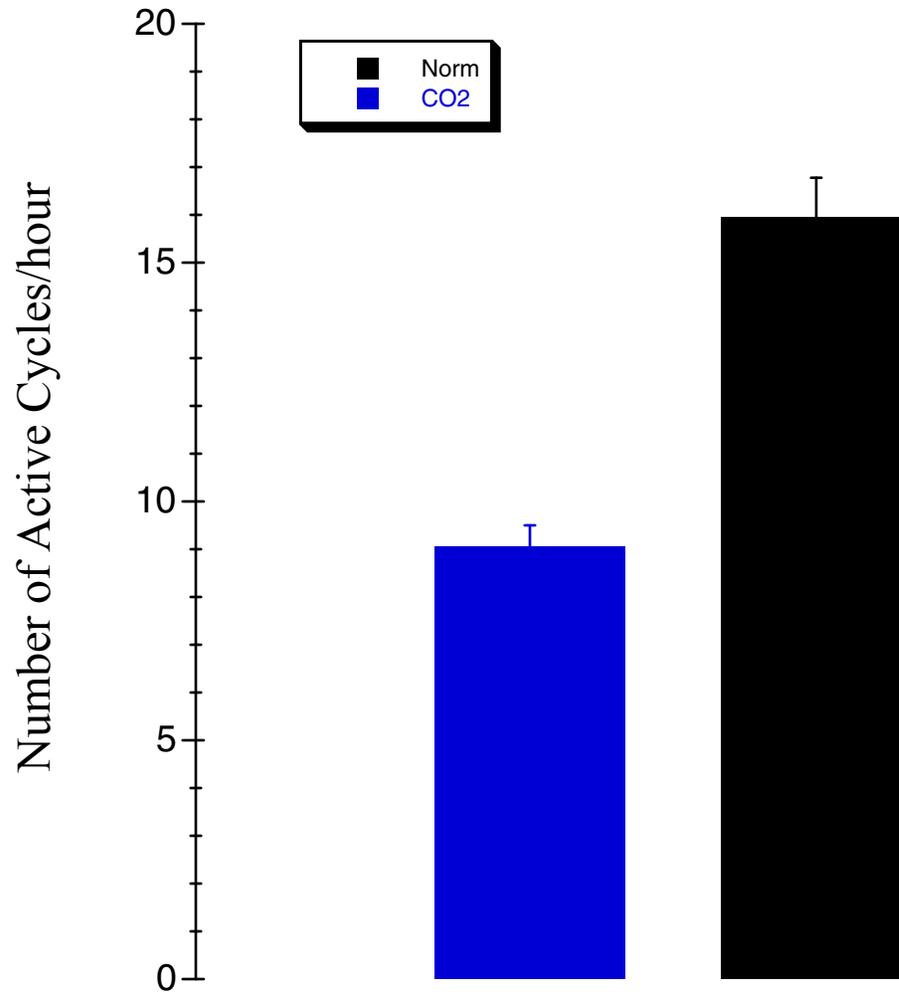
Most pronounced at high activity levels



# Metabolism is reduced at 0.1% CO<sub>2</sub> and 20° C



# Activity is reduced at 0.1% CO<sub>2</sub>



*The synergistic effect of these three climate-related variables ( $O_2$ ,  $T^\circ C$  and  $CO_2$ ) may be to vertically-compress the habitable nighttime depth range of the species*

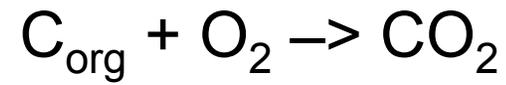
*-Rosa and Seibel, 2009 PNAS*



QuickTime™ and a  
decompressor  
are needed to see this picture.

QuickTime™ and a  
decompressor  
are needed to see this picture.

QuickTime™ and a  
decompressor  
are needed to see this picture



$$\Delta G = \Delta G^\circ * \ln \left\{ \frac{fCO_2}{(C_{\text{org}})(fO_2)} \right\}$$

$$\text{Log}_{10} (pO_2/pCO_2) = \text{Respiration Index}$$

.re.

*“A simple numerical constraint linearly related to available energy”*

QuickTime™ and a  
decompressor  
are needed to see this picture

## Two faulty assumptions:

- 1) That the respiration equation proceeds toward Equilibrium as if in a closed system.
- 2) That gas partial pressures inside cells resemble Those in the environment.

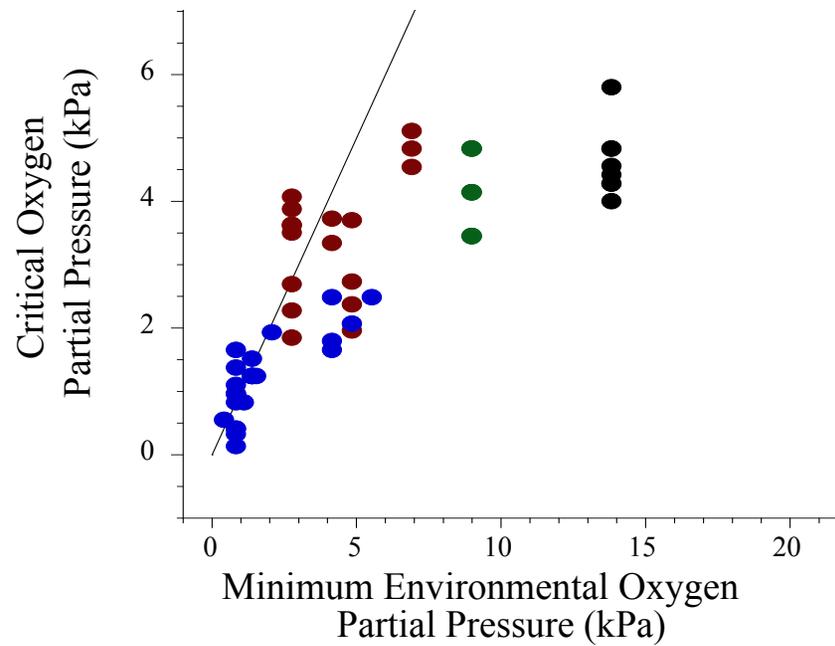
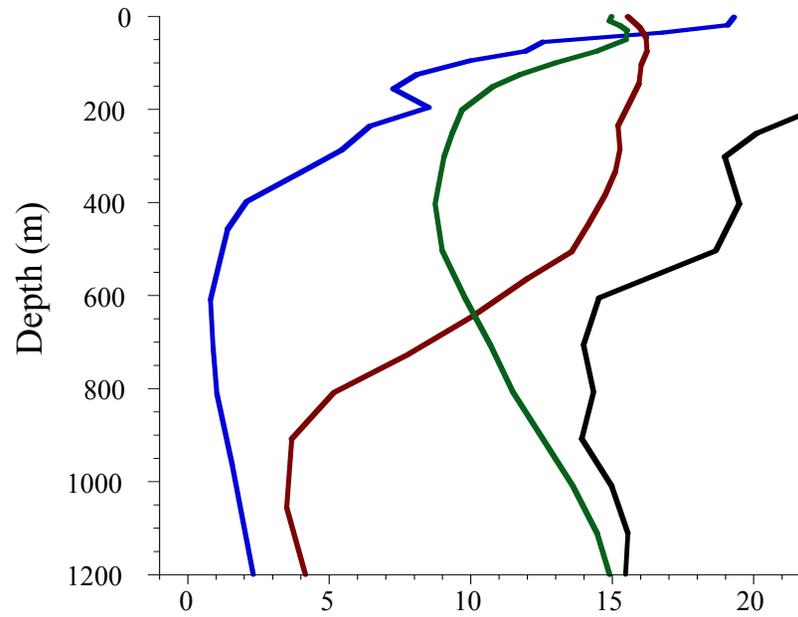
QuickTime™ and a  
decompressor  
are needed to see this picture

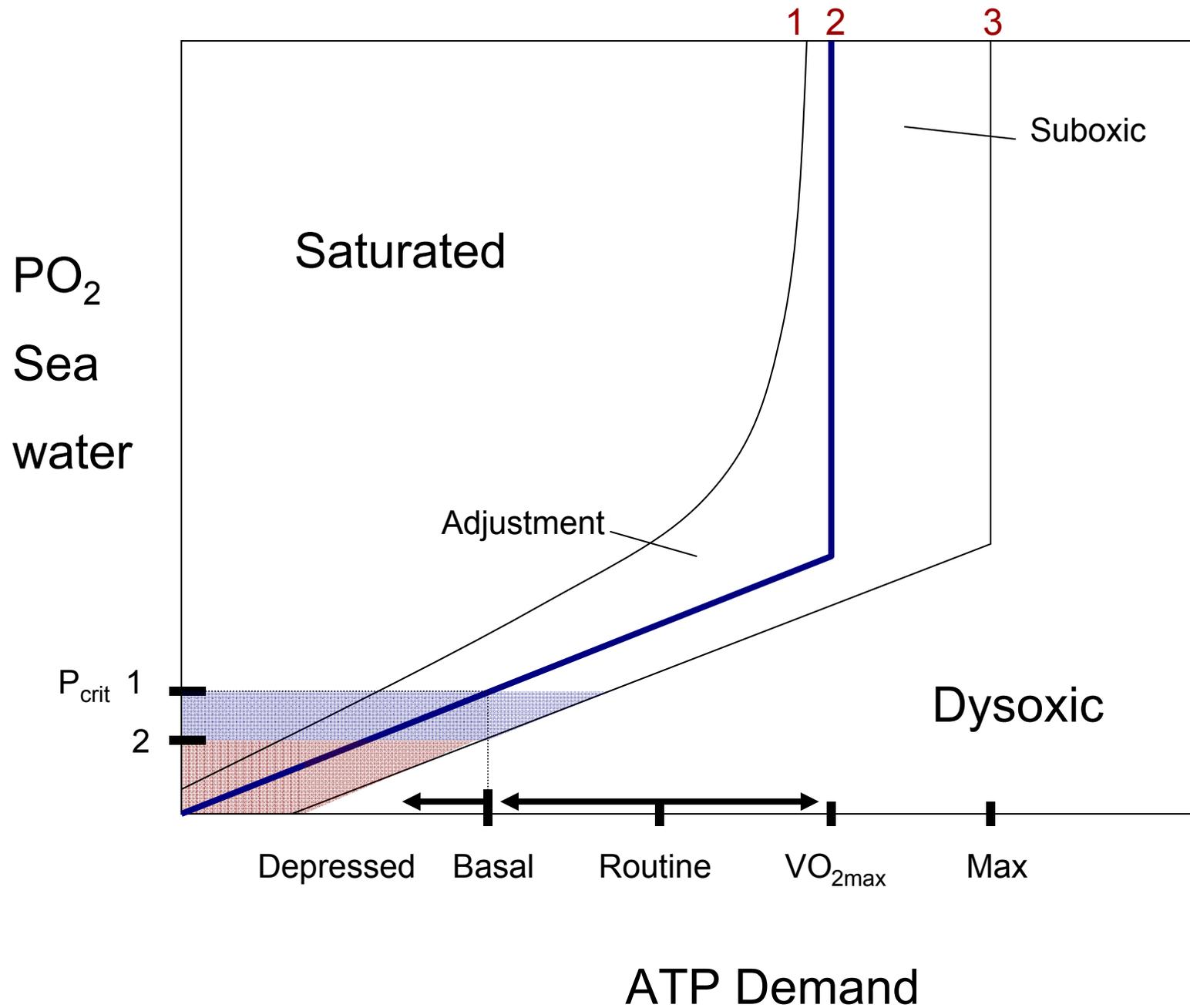
## Leading to faulty conclusions:

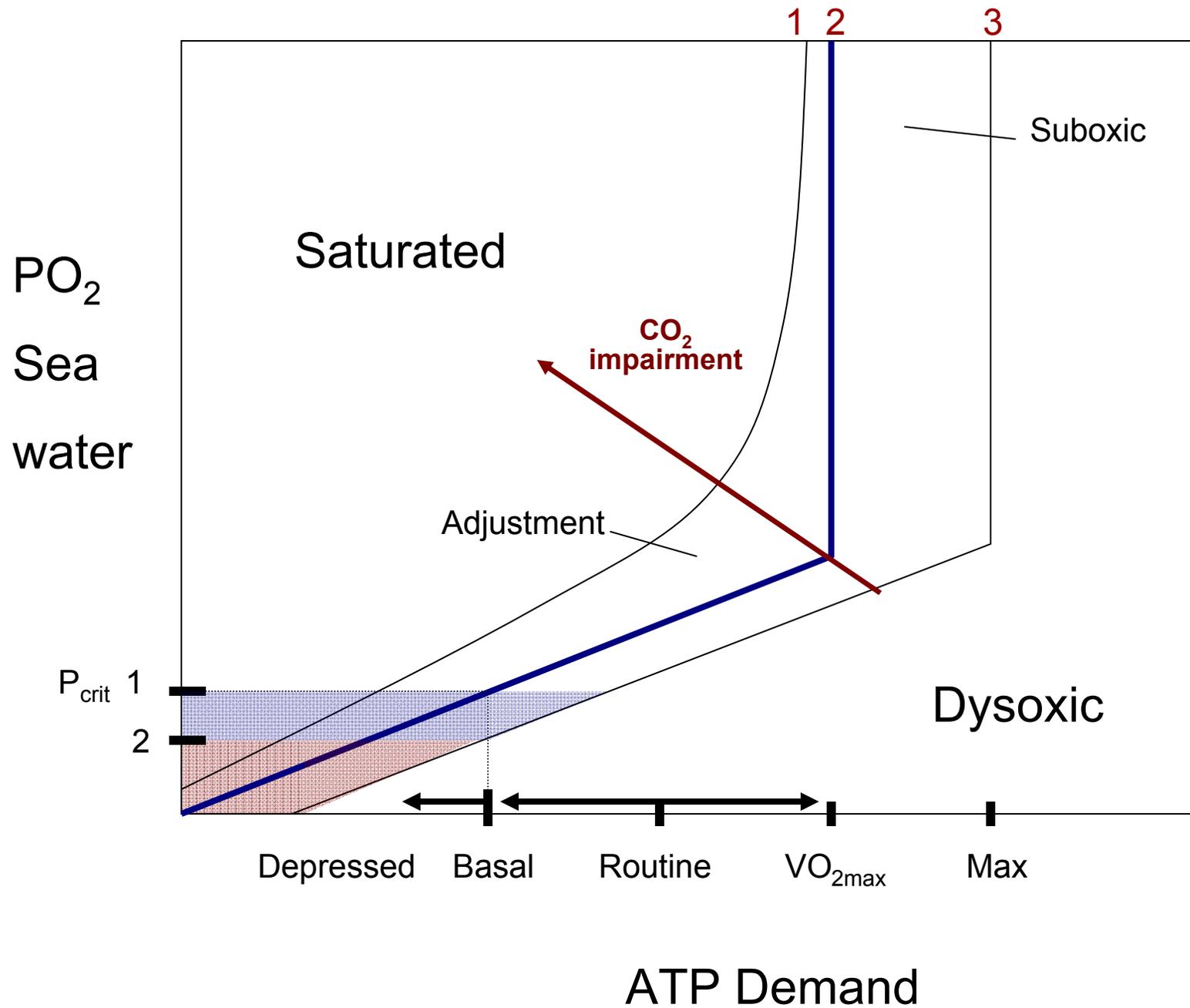
*“For the vast areas of the ocean that are well-oxygenated, the rise in oceanic CO<sub>2</sub> concentrations will exert a negligible effect on the normal aerobic functioning of adult marine animals”.*

*“Even if oxygen levels do not decline, the oceanic dead zones will expand as a result of rising CO<sub>2</sub> concentrations”.*

# Evolved response to environmental oxygen







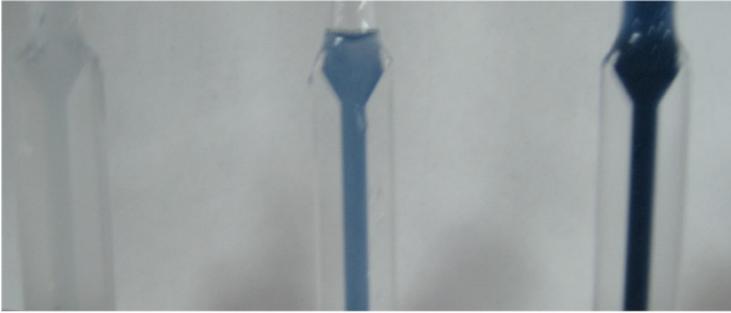
# **Ocean Acidification Research**

**What is the goal?**

**To determine critical CO<sub>2</sub> levels?**

**To demonstrate that CO<sub>2</sub> is a problem?**

**Funding to investigate interesting questions?**



*Dosidicus* has high oxygen affinity

QuickTime™ and a  
decompressor  
are needed to see this picture.



# Calcification

Controlled by  $[\text{CO}_3^{=}]$ ? Or too simplistic?

QuickTime™ and a  
decompressor  
are needed to see this picture.